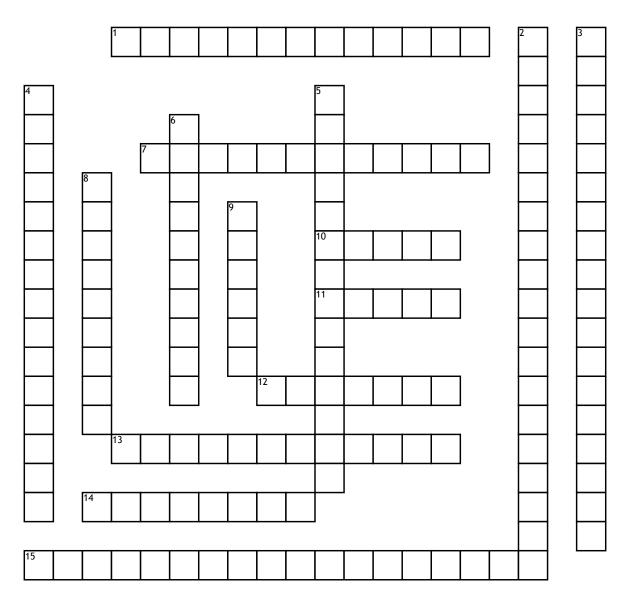
Name:	Date:
-------	-------

## 7th Grade Contract



## **Across**

- 1. a table filled with elements
- 7. The number of protons in an atom; also identifies the element.
- **10.** Vertical columns of elements with similar properties, numbered 1-18
- **11.** Good conductors of heat & electricity
- **12.** substance made up of atoms with the same identity

- **13.** The most reactive of all metals; don't occur in nature in their element form
- **14.** atoms of two or more elements combined
- **15.** Most familiar metals because they often occur in nature uncombined

## Down

- **2.** Shiny, ductile, and malleable; combine readily with other elements
- **3.** Uses the element symbol and dots to represent outer energy level

- **4.** Devised first periodic table based on atomic mass
- **5.** Arranged elements by atomic number and is what we use today
- **6.** Composed mostly of protons and neutrons in the nucleus; also known as mass number
- **8.** type of matter with a fixed composition that CANNOT be separated by physical means
- **9.** Horizontal row of elements that contain increasing numbers of protons & electrons