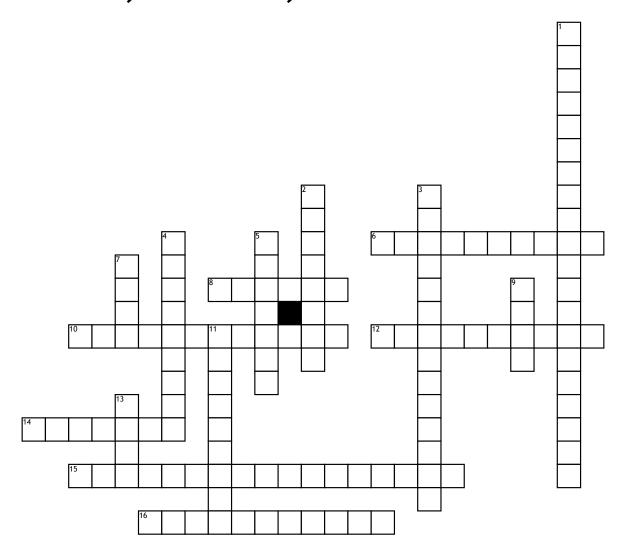
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## Acids, Bases, and Solutions



## **Across**

- **6.** The measure of how well a solute can dissolve in a solvent at a given temperature.
- **8.** Something that gets dissolved in another substance. An example of this is salt.
- **10.** A negatively charged ion made of hydrogen and oxygen.
- **12.** What is a mixture that can be easily seperated?
- **14.** A range of numbers used to express the concentration of hydrogen atoms in a solution

- **15.** A mixture that has too much solute that it can't disslove.
- **16.** A positevely charged ion formed of a hydrogen atom that has lost its electron.

## **Down**

- 1. A mixture that has a lot of solute dissolved in it.
- 2. A well-mixed mixture.
- **3.** A mixture that has little solute dissolved in it.
- **4.** The way in which acids react with metals so as to eat away at the metal.

- **5.** A substance dissolves in this. An exampple of this is water.
- **7.** What causes litmus paper to turn red?
- **9.** What causes litmus paper to turn blue?
- 11. A compound that changes color in the presence of an acid or a base.
- **13.** What is an ionic compound made from the neutralization of an acid and a base?