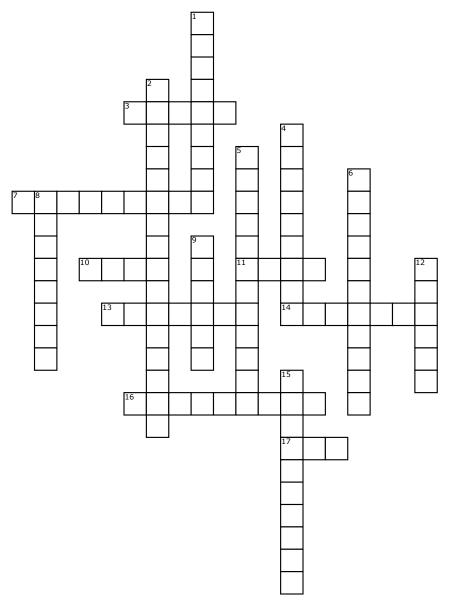
## Acids and Alkalies



## Across

- **3.** Can react with an acid and can be reactive or not
- 7. Loss of oxygen
- **10.** The product of a chemical reaction
- **11.** Measures from 1 to 6 on the PH scale and can be very reactive
- **13.** Used to measure the strength of an acid or an alkali
- **14.** Neither an acid or an alkali

- **16.** Gain of electrons
- **17.** Result of gain or loss of electrons

## **Down**

- 1. Gaining of oxygen
- **2.** Order in which the metals are placed due to strength
- 4. Loss of electrons
- **5.** The removal of one metal to be replaced by a higher reacting one.

- **6.** A stronger form of an acid and has not mixed with much water
- **8.** Can make an ion or an atom
- **9.** Measures from 8 to 14 on the PH scale and can neutralise an acid
- **12.** A weaker form of an acid and has mixed with lots of water
- **15.** Can vary in stages but is when acids dissolve in water