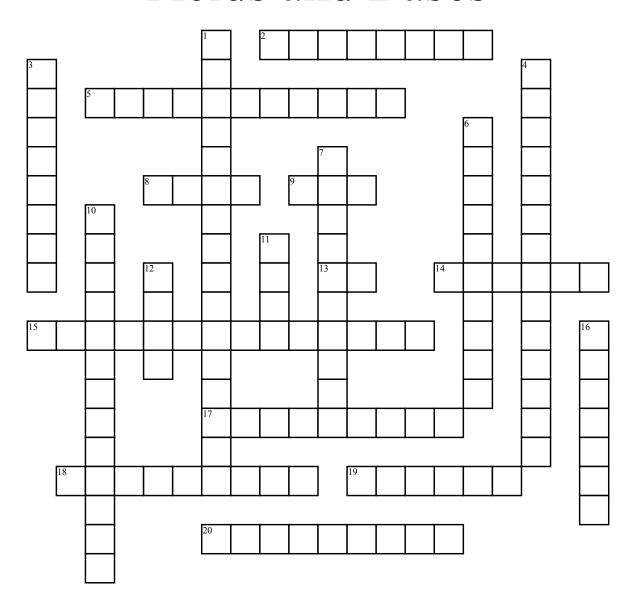
Name:	Date:	

Acids and Bases



Across

- **2.** the point in a titration that the substance fully changes color
- **5.** an indicator that turns different colors depending on pH
- **8.** substances that dont fully ionize
- **9.** a scale to determine hydroxide ions in a base
- **13.** measured in a range from 1-14
- **14.** kinds of acids like Hcl and H2SO4

- **15.** said that acids are proton donors
- 17. an example is litmus paper
- **18.** an experiment that determines the concentration of an acid or a base
- **19.** a substance that resist a change in pH
- **20.** said that an acid had hydronium ions and bases had hydroxide ions

Down

1. the point when the solute and the solvent become equal

- **3.** a substance with 2 protons
- **4.** the way number 15 is displayed
- **6.** a substance that can either be an acid or a base
- 7. only has one proton
- **10.** acids have a high concentration of these
- **11.** a pH below 7
- 12. has a pH above 7
- **16.** an element with a pH of 7 is considered this