

Name: _____

Date: _____

Period: _____

Acids and Bases

V K J W W N O I T U L O S L A R T U E N S P X A
Z R I V I T I A U E E H U J C M Z Y E S F B N V
U L R V A Y A F B A F J W Q H K S Y U L L R N K
B N O I T C A E R N O I T A Z I L A R T U E N N
J F E V Z W Q Z U F V J S L L Z K M S Z W J N O
B R O N S T E D L O W R Y A C I D P I B G L D I
A L T M U I M U Z N Y A D C Y C V H J Q F A N M
N Y R O N N O I E D I X O R D Y H O N C A C H U
S O S T R O N G A C I D X E O T P T C K R N W I
T B N K W E A K B A S E L E H L M E Y N R R E N
R B E S A B Y R W O L D E T S N O R B H H A A O
O L R S D R Z L T N L T X N P K Z I B P E J K R
N R S G F L Y X E R R E F F U B N C R R N F A D
G K N O I T A I C O S S I D V Z M L J S I T C Y
B R D I C A S U I N E H R R A B P D E Z U Y I H
A K J A V O X M L C O N J U G A T E B A S E D K
S M X M W W E I B A U I R U S D Y C K W B Q A C
E Q J N O I T A Z I N O I F L E S H O P A Q S T
A C I D I C S O L U T I O N F E P K C Y S D S L
Y T N A T S N O C D E C U D O R P N O I E O Y W
F R Z L C O N J U G A T E A C I D A Y N W S S D
R P O V C F H S V Q M B N T M N B V Q T Z O U X
D D J D Q Y H S B A S I C S O L U T I O N Q F Y
D N I V W P H S C A L E Q K J W K X T V V E T T

- neutralization reaction
- Ion produced constant
- Bronsted-Lowry Acid
- Bronsted-Lowry Base
- neutral solution
- acidic solution
- Self-Ionization
- Arrhenius Acid
- Arrhenius Base
- basic solution
- conjugate acid
- conjugate base
- hydronium ion
- hydroxide ion
- Dissociation
- strong acid
- strong base
- amphoteric
- weak acid
- Weak Base
- ph scale
- Buffer

