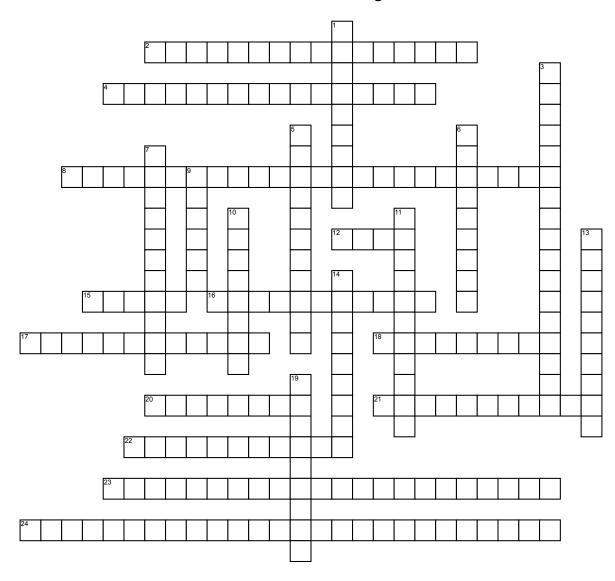
## **Aromatic Chemistry Crossword**



## Across

- 2. Acts as a "Road Block" for resonance forms to be created
- **4.** Groups that tend to increase the rate of electrophillic aromatic substitution
- **8.** What kind of reaction does Pyridones undergo with POCL3
- **12.** The name given to a 1,4-disubstituted benzene
- **15.** A heterocylic aromatic compound consisting of 4 carbon atoms and 1 oxygen atom
- **16.** Overlap of more than 2p orbitals in an acyclic or cyclic system
- 17. A reduction reaction that uses BASIC conditions to convert an aldehyde or ketone to an alkane
- **18.** The name of the reaction where the diazonium ion reacts with cuprous salt

- **20.** a very unreactive aromatic imine and foul smelling
- **21.** What compound is created when benzene diazonium chloride reacts with KI at a 25-100°C
- **22.** Used to show the movement of an electron pair
- 23. A reaction which invloves the addition of an acyl group to an aromatic compound
- **24.** The mechanism for the Nitration of Benzene

## **Down**

- **1.** The electrophile/ion that is produced during the nitration of benzene
- **3.** The catalyst needed for the acylation of an aromatic compound
- **5.** The number rule that will estimate whether a planar, cyclic and conjugated compound will exhibit aromatic properties

- **6.** Two molecules that have the same molecular formula but have a different structural formula
- 7. The product produced in the oxidation of aromatic alkenes (except tertiary alkenes!) with KMNO4
- **9.** Any compound that has a hydroxyl group which is directly attached to the benzene ring
- **10.** Compounds that can undergo polybromination with the addition of Br2
- 11. The postively charged ion created during electrophillic aromatic substitution
- **13.** A reduction reaction that uses ACIDIC conditions to convert an aldehyde or ketone to an alkane
- **14.** The reagant needed to create Thiophene from hexane-2,5-dione
- **19.** The component that allows for several Lewis structures to be formed through the delocalisation of electrons