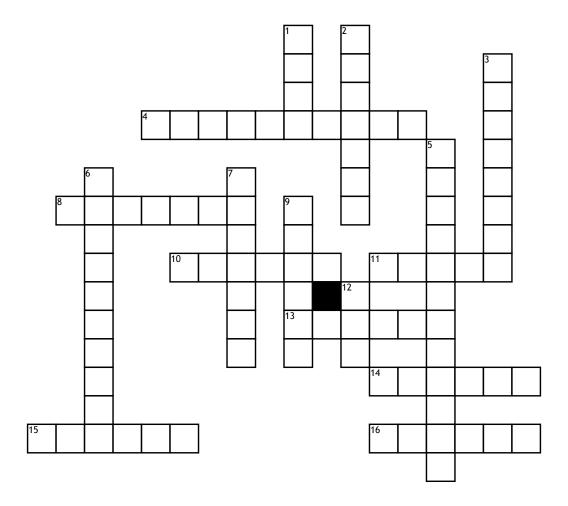
## Atomic Structure and Periodic Table



## <u>Across</u>

- **4.** ancient Greek who first thought of atoms
- **8.** electrons in the outermost electron shell
- **10.** a positively charged ion
- 11. a negatively charged ion
- **13.** positively charged particle in the nucleus
- **14.** the shortened form of an elements name; found on the periodic table

- **15.** a horizontal row on the periodic table
- **16.** the rings around the nucleus in a Bohr model

## Down

- 1. last name of man whose model shows electrons in shells around the nucleus
- **2.** same number of protons with different number of neutrons
- **3.** negatively charged particles that fly around the nucleus
- **5.** represents the number of protons an atom has
- **6.** protons plus the number of neutrons
- **7.** particle located in the nucleus with no charge
- **9.** the vertical columns on the periodic table
- 12. any charged particle

## **Word Bank**

| groups     | cation | valence | symbol       |
|------------|--------|---------|--------------|
| Democritus | Bohr   | proton  | massnumber   |
| electron   | shells | period  | neutron      |
| anion      | ion    | isotope | atomicnumber |