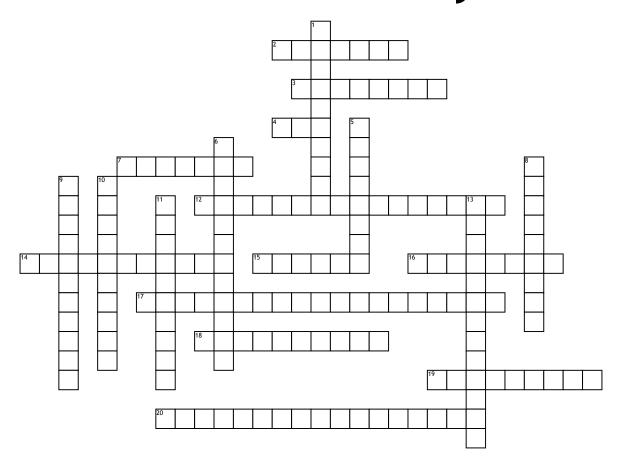
Atomic Theory



<u>Across</u>

- **2.** Subatomic particle with no charge
- 3. Discovered the neutron
- **4.** One of the first four elements
- 7. Where protons are located
- **12.** The theory the the shape of an atom was consistent throughout with electrons mixed in
- **14.** Came up with Quantum Mechanical Model
- **15.** Subatomic particle with a positive charge

- **16.** A subatomic particle with a negative charge
- **17.** The atomic theory we use today
- **18.** Came up with the uncertainty principal
- 19. Bohr's model of an atom
- **20.** Dalton's model of an atom **Down**
- 1. Used Gold Foil Experiment to disprove plum pudding model
- **5.** Did the oil drop experiment
- **6.** Rutherfords model of an atom

- **8.** Discredited the idea of the atom by Leucippus and Deocritus
- 9. How many times the nucleus is smaller than the atom
- **10.** Used a high vacuum cathode-raytube in his experiment
- 11. Came up with the idea that all matter is made of atoms
- **13.** Where electrons are located