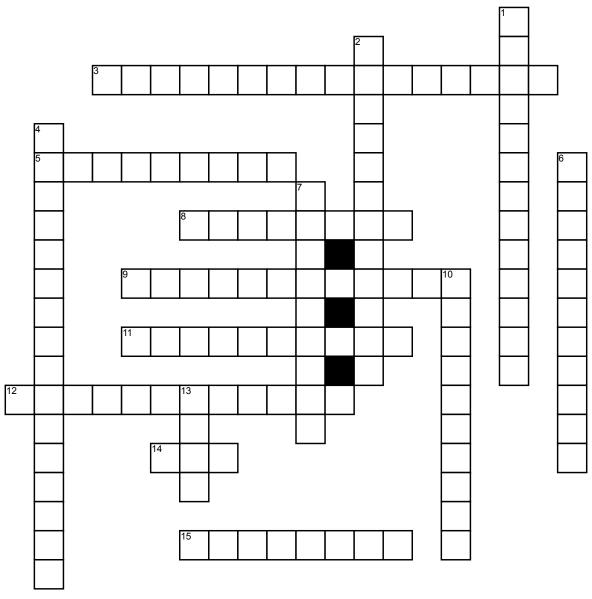
Atoms and molecules



Across

- **3.** Molecule that shares electrons equally.
- **5.** When two opposite charges of ions are attracted to each other
- **8.** Substance formed when two or more chemical elements are bonded together.
- **9.** A chemical force holding atoms together.
- **11.** Two atoms sharing three pairs of eletrons.

- **12.** The equal sharing of electrons between atoms
- **14.** Positively or negatively charged atom or molecule.
- **15.** When a molecule has one positively charged end and one negatively charged end.

Down

- **1.** A molecule that has unequal electron sharing.
- **2.** The sharing of eletrons between atoms.

- **4.** Molecules that are only composed of two atoms.
- **6.** The lowest ratio possible of an ionic compound.
- 7. When electrons are not shared equally between atoms.
- **10.** When two electron pairs are shared by two atoms.
- **13.** The smallest possible amount of a chemical element.