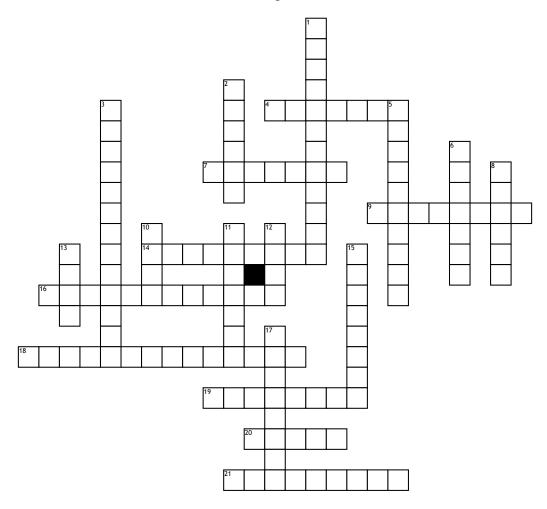
## **Biochemistry Crossword**



## Across

- 4. What is a polymer consisting of one or more long chains of amino acids
- **7.** What is a substance in which a solute is dissolved to form a solution
- **9.** What are two or more atoms bonded together
- **14.** What is the the monomers (building blocks) of proteins
- **16.** What is a chemical bond formed due to one or more shared pairs of electrons
- **18.** What are many sugar molecules bonded together (polymer)
- 19. What is the property that causes molecules to stick to each other

- **20.** What is an organic compound stores the most energy and is used to form cell membranes; formed from fatty acids and a glycerol
- 21. When one atom steals electrons from another Down
- 1. When a positively charged H atom of one molecule is attracted to a negatively charged O or N atom of another
- **2.** What is a positively charged subatomic particle
- **3.** What is the organic compound used as our key source of energy, made of monosaccharides.
- ${\bf 5.}$  What is the monomer used to build nucleic acids (5 types)
- **6.** What is the measurement system used to indicate the concentration of hydrogen ions (H+) in solution (acid or bas); ranges from 0 to 14

- **8.** What is a substance that is dissolved in a solvent to make a solution
- **10.** pH>7; substance that releases hydroxide (OH-) ions when dissolved in water
- **11.** What is the sub units in a complete molecule, they form polymers
- **12.** ph <7; releases H+ ions when dissolved in water
- **13.** What is a VERY small particle made of protons (+), electrons (-) and neutrons (no charge)
- **15.** What is a negatively charged subatomic particle
- 17. What is the attraction of different substances (sticking together like with glue)

## **Word Bank**

ionic bond Amino acid atom base lipid protein nucleotide molecule electron pH Scale proton adhesion cohesion Solvent Carbohydrates hydrogen bond acid polysaccharide Solute covalent bond Monomer