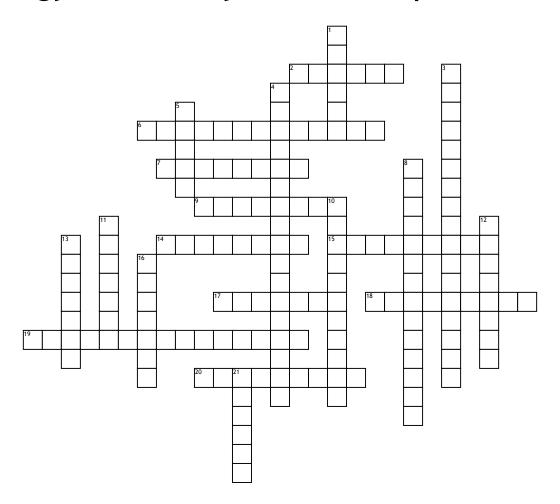
Name: Date:

Biology Chemistry Terms Chapter 2.1-2.5



Across

- 2. acquires a net positive charge
- 6. energy in action
- 7. moles per liter
- **9.** heterogenes mixtures, which means that their compositions is dissimilar in different areas is the mixture
- **14.** are substances compound of two ir more components physically intermized
- **15.** homogeneous mixtures of components that may be gases, liquids and solids
- 17. neutral charge

- **18.** a chemical bond between atoms formed by the transfer of one or more electrons from one atom to the other
- **19.** most atoms with only one or two valence shell electrons
- 20. bear a negative charge

Down

- 1. anything that accupies space and has mass
- **3.** the molecules formed are electrically balanced
- **4.** electron hungry and attract very strongly, a capability
- 5. acquires a net negative charge

- **8.** electrons forming the electron cloud around the nucleus of an atom occupy regions of space
- **10.** are heterogeneous mixtures with large, pften visible solutes that tend to settle out
- **11.** substance present in the greatest amount
- **12.** known elements have two or more structural variations
- 13. substances present in smaller amounts
- 16. positive electrical charge
- 21. capacity to do work

Word Bank

Neutron

Matter **Electrons** Anion Kinetic energy Solutions Colloids Nonpolar Molecules Electronegativity Cation Solutes Electron shells Ionic bond Electropositive **Protons** Energy Solvent Mixtures Suspensions Molarity Isotopes