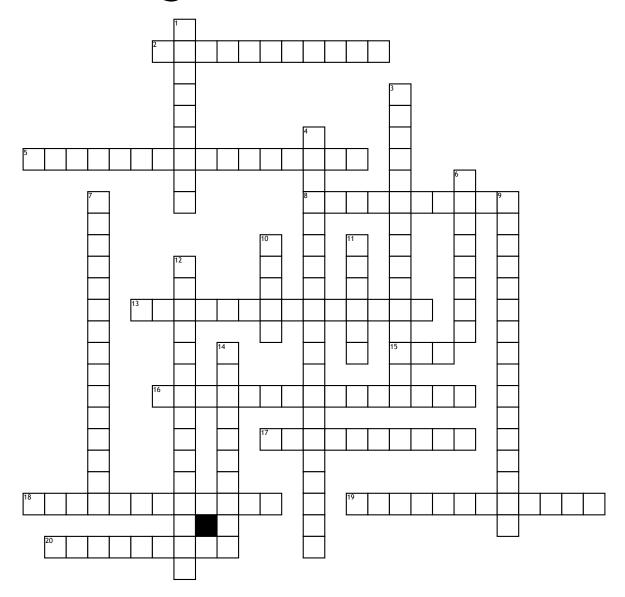
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## Bonding & Reactions Unit 6



## **Across**

- **2.** Notations for a positive or negative ionic charge
- **5.** The energy required to remove the electron from its shell to an isolated atom
- **8.** A chemical bond between two atoms
- 13. Many atoms with a charge
- **15.** Electrons that are gained or lost
- **16.** A diagram that shows bonding between atoms
- **17.** A chemical bond between two chemical elements

- **18.** The sharing of electrons to obey its octet rule
- **19.** An electrical force that holds atoms together
- **20.** A bond that occurs between a metal and nonmetal **Down**
- 1. Number of atoms
- **3.** The type of chemical bond between atoms in a metallic element
- **4.** When atoms are spatially arranged
- 6. A covalent compound

- **7.** An expression that states the number and type of atoms
- **9.** Molecule composed into two atoms
- 10. Negatively charged ion
- 11. Positively charged ion
- **12.** An electron located in the outermost shell
- **14.** A chemical bond where three atoms share three pairs of electrons