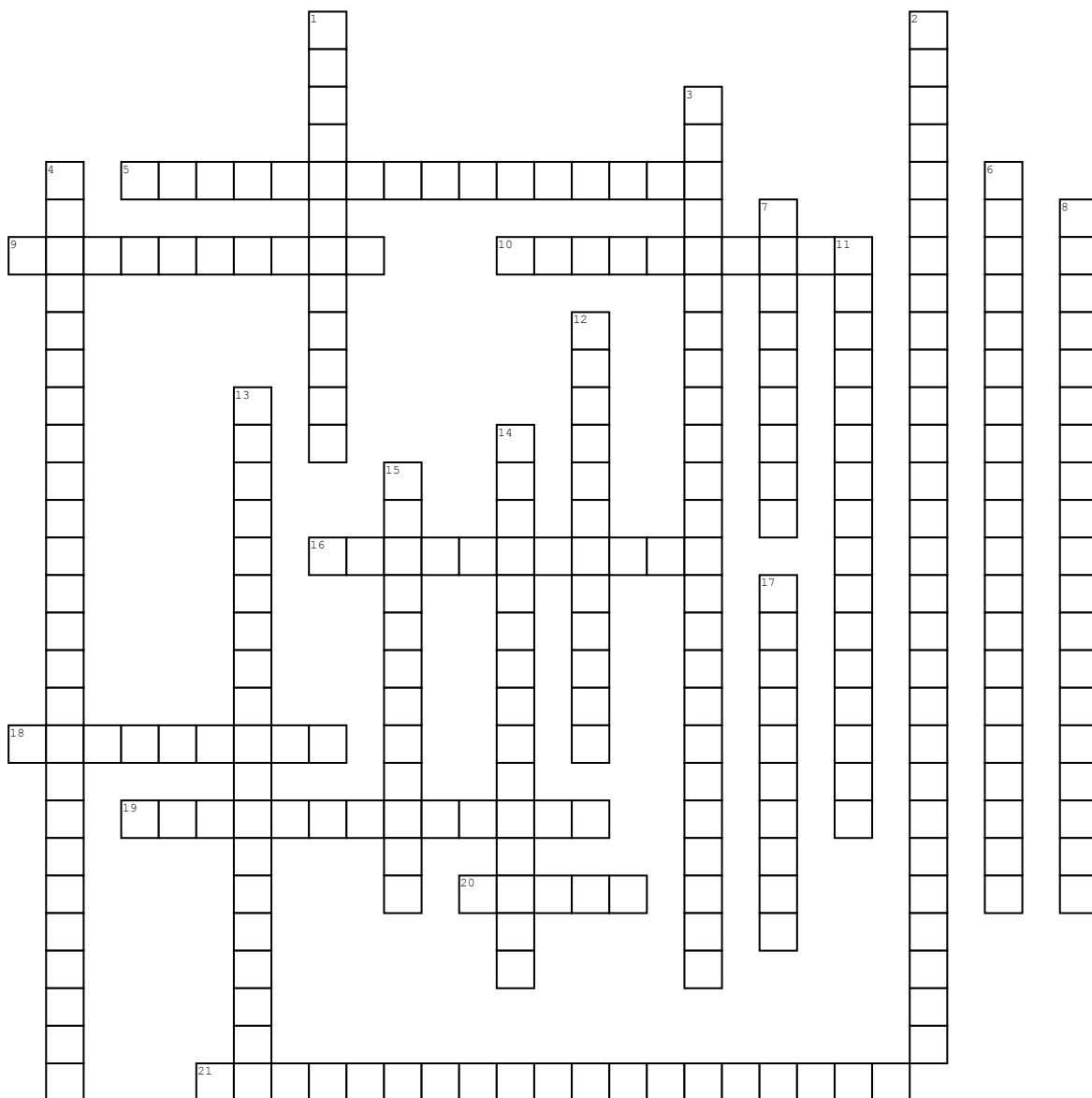


Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

# CHEMISTRY CH 10 REVIEW



## Across

5. Where can you find molar mass
9. mass of the compound AgNO2
10. multiply across
16. How many grams are in 7.14 times  $10^{22}$  of H2O
18. mass of the compound NaCl
19. Avogadro's number
20. molar mass of ammonium sulfate
21. how many sig figs in 106

## Down

1. how many grams are in 2 mole of NaCl
2. How many molecules are in 4.95 mole of Na3PO4
3. The formula for finding percent composition?
4. The percent composition of Boron acetate
6. if you know the number of moles of compound?
7. what does one need to know in order to convert from grams to moles?

8. How many sig figs in 0.053
11. Percent Composition of Sodium Bromide
12. How many molecules are in 2.00 moles of H2O
13. How many sig figs in 1.50
14. how many atoms are in 0.250 moles of Rb
15. How many moles are in  $2.91 \times 10^{22}$  atoms of He
17. Mass of the compound KBr