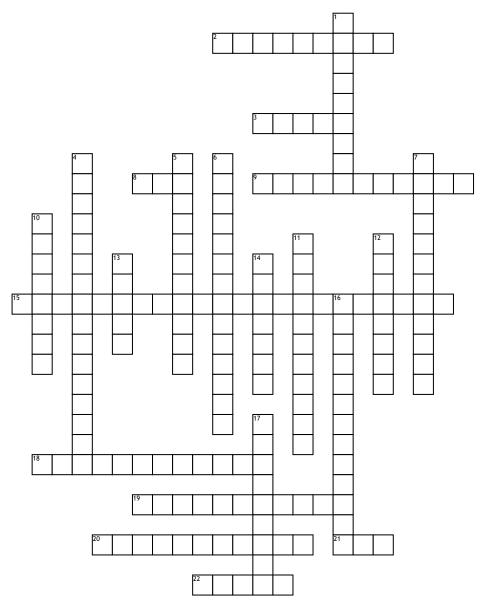
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Carbonyl Chemistry



Across

- **2.** Principle of bonding between multiple structures
- 3. Which molecule destroys the organometallic reagent?
- **8.** How many carbonyl stretches to anhydrides usually give?
- **9.** The carboxylate anion can act as a...
- **15.** Compounds which contain carbon metal bonds
- 18. Simplest Carbon compound
- **19.** Reduces double bond character of a carbonyl group
- 20. Strong bond, hard to stretch and takes up more energy

- **21.** type of hybridisation in a carbonyl group
- **22.** Acid + alcohol produces **Down**
- **1.** Two acids without a water molecule
- **4.** An increase in energy to excite the vibration causes...
- **5.** Reducing the double bond is known as...
- **6.** It is sp2 hybridised and molecule is flat
- **7.** Widely used to methylate carboxylic acids and form esters
- **10.** A physical property of carbonyl compounds

- **11.** Molecules that have four electron groups around their central atom
- **12.** What complementary H-bonding sites of carboxylic acids do in solution?
- **13.** Spectroscopy of carboxylic acids O-H stretch is
- **14.** Reagent used to perform addition reactions to carbonyls
- **16.** Differ from acids by having a leaving group in place of the OH functionality
- **17.** Amide with a boiling point of 221.1 C