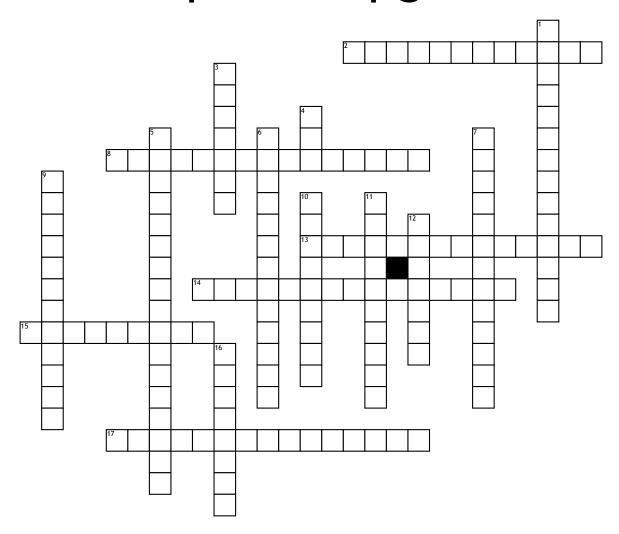
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## **Across**

- **2.** Bond between atoms made by sharing electrons
- **8.** Combining capacity of an element
- **13.** A compound with only 2 elements
- **14.** A group of chemical symbols explaining atoms in a compounds ratio
- **15.** Without water
- **17.** Electron paired with an atom that can be in a chemical bond

## Down

- 1. Positively charged molecule on 1 end and negative on the other
- **3.** Material with repeating geometric form by element or compound
- **4.** Charged particle from an atom with gained or lost electrons
- **5.** Particle made of 2 covalently bonded atoms
- **6.** Group of covalently bonded atoms forming 1 charge

- **7.** Big element formed by covalent bonding
- **9.** Bond where ions are held together by attraction from opposite charges
- **10.** Small number giving a ratio of atoms of each element combine in a compound
- 11. Results of forces from attraction between positively charged nuclei and electrons
- 12. A crystal with water
- **16.** Neutral particle made by atoms bonded covalently