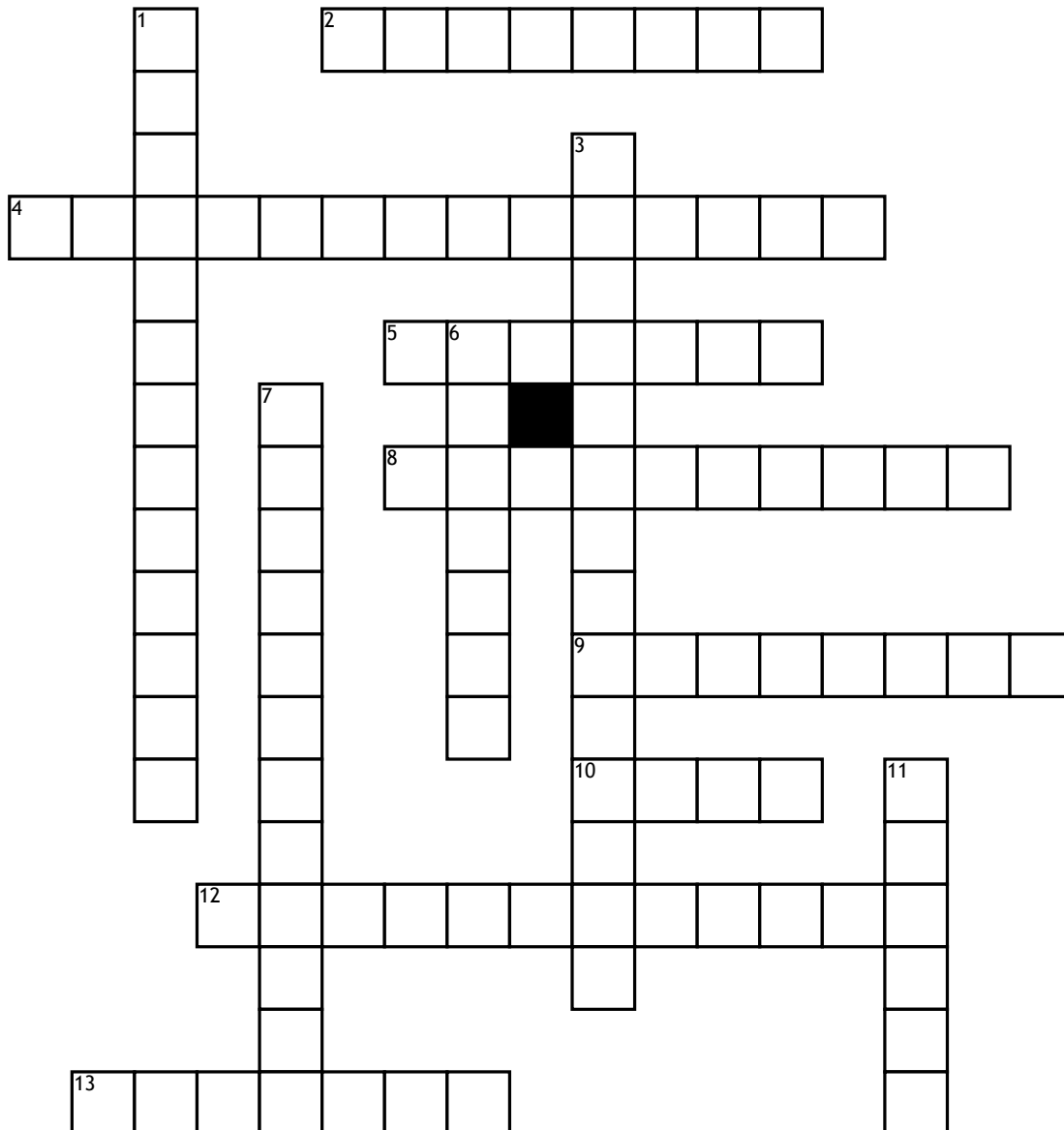


Chapter 2: Matter - Claire Avery



Across

2. A neutral group of atoms held together by covalent bonds
 4. A change that occurs when one or more substances change into interlay new substances with different properties
 5. Measure of how much matter is contained in a certain volume of a substance
 8. The capacity of a substance to combine chemically with another substance

9. Substance that is made of two or more different elements joined by chemical bonds
 10. Smallest unit of an element that maintains the property of that element
 12. Temperature at which the equilibrium vapor pressure of a liquid equals the atmospheric pressure and liquid becomes a gas
 13. Combination of substances that are not chemically combined

Down

1. Matter that has a fixed composition and definite properties
 3. A change of matter from one form to another without a change in chemical properties
 6. A pure substance made of only one kind of atom
 7. Temperature and pressure at which a solid becomes a liquid
 11. Anything that has mass and takes up space