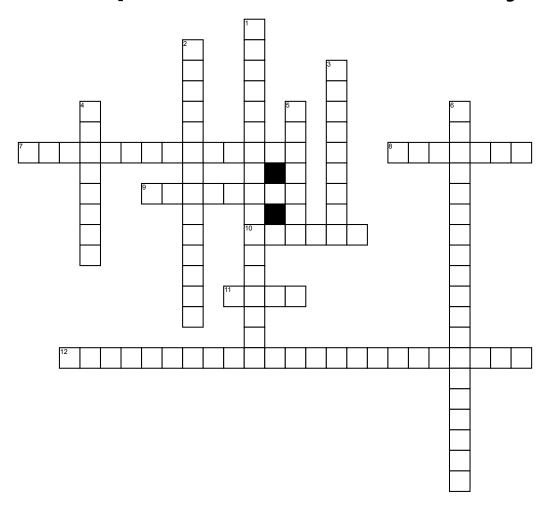
## Chapter 2. Vocabulary



## **Across**

- 7. a change in matter in which the substances that make up the matter change into other substances with different physical and chemical properties.
- **8.** matter that can vary in composition.
- **9.** two or more atoms that are held together by chemical bonds and act as a unit.
- **10.** everything that has mass and takes up space.

## **Word Bank**

physical change heterogenous mixture atom substance

- **11.** small particle that is the building block of all matter.
- **12.** states that the total mass of a chemical reaction is the same as the total mass after the chemical reaction.

## **Down**

- 1. a mixture in which two or more substance are evenly mixed but no bonded together.
- **2.** a change in size, shape, form or state of matter that does not change the matter's identity.

- **3.** matter with a composition that is always the same.
- **4.** substance made of two or more elements that are specifically joined in a specific combination.
- **5.** a pure substance made up of only one kind of atom.
- **6.** mixture in which the substances are not evenly mixed.

homogenous mixture compound matter molecule chemical change mixture law of conservation of mass element