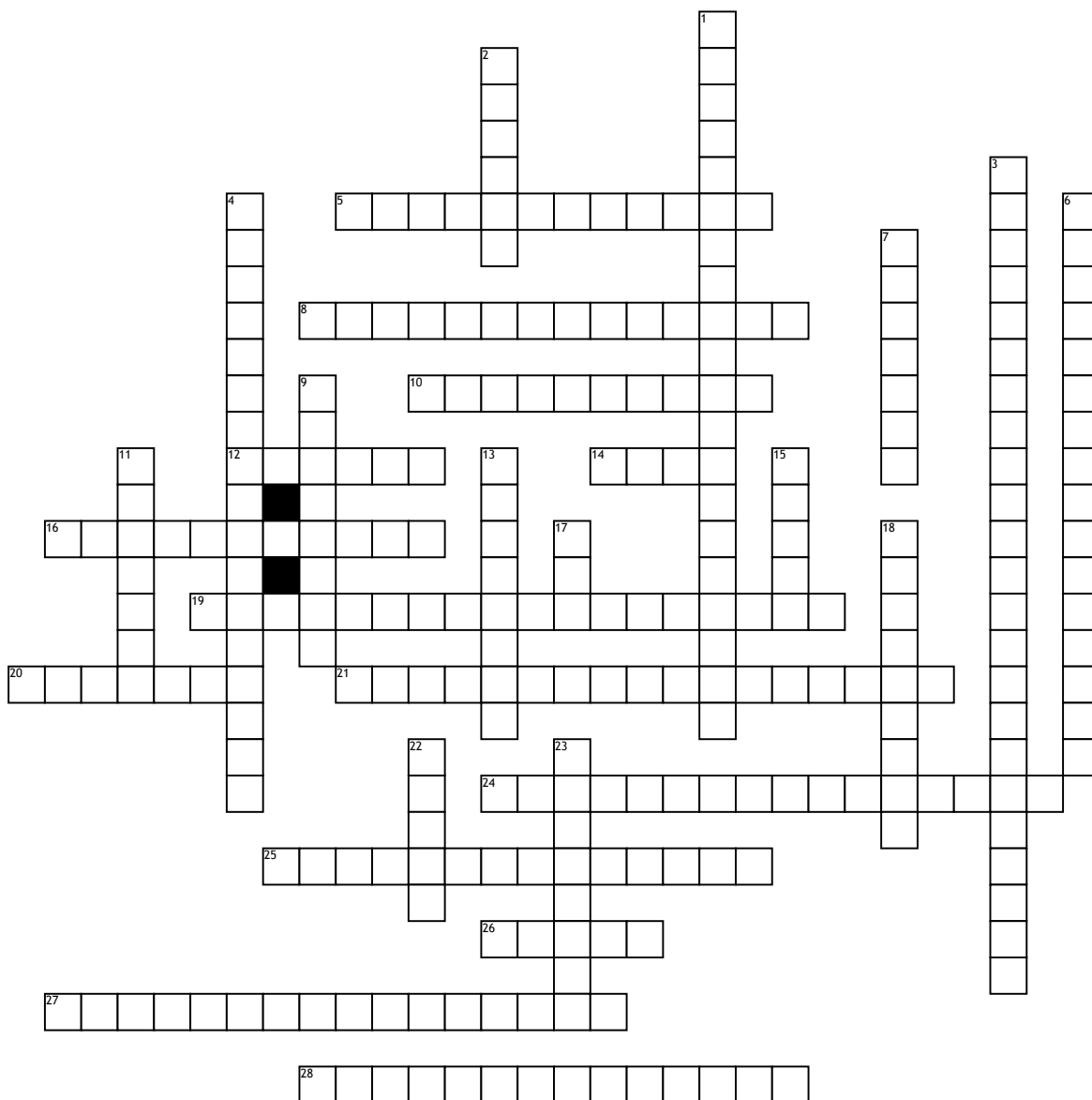


Name: _____

Date: _____

Chapter 2 vocab



Across

5. a liquid is boiled to produce a vapor that is then condensed into a liquid
 8. a change that produces matter with a different composition than the original matter
 10. the process that separates a solid from a liquid in a heterogeneous mixture
 12. a measure of the space occupied by the object
 14. a measure of the amount of matter the object contains
 16. a solid that forms and settles out of a liquid mixture
 19. a mixture in which the composition is uniform throughout
 20. a physical blend of two or more components
 21. a property that depends on the type of matter in a sample, not the amount of matter

24. one or more substances change into one or more new substances
 25. some properties of a material change but the composition of the material does not change
 26. used to describe any part of a sample with uniform composition and properties
 27. the ability of a substance to undergo a specific chemical change
 28. each element is represented by a one or two letter chemical symbol

Down

1. a mixture in which the composition is not uniform throughout
 2. a form of matter that has an indefinite shape flows yet has a fixed volume
 3. states that in any physical change or chemical reaction mass is conserved
 4. a property that depends on the amount of matter in a sample

6. a quality or condition of a substance that can be observed or measured without changing the substance's composition
 7. a substance produced in the reaction
 9. another name for homogeneous mixture
 11. the simplest form of matter that has a unique set of properties
 13. a substance that contains two or more elements chemically combined in a fixed proportion
 15. the gaseous state of a substance that is generally a liquid or solid at room temperature
 17. a form of matter that takes both the shape and volume of its container
 18. matter that has a uniform and definite composition
 22. a form of matter that has a definite shape and volume
 23. a substance present at the start of the reaction