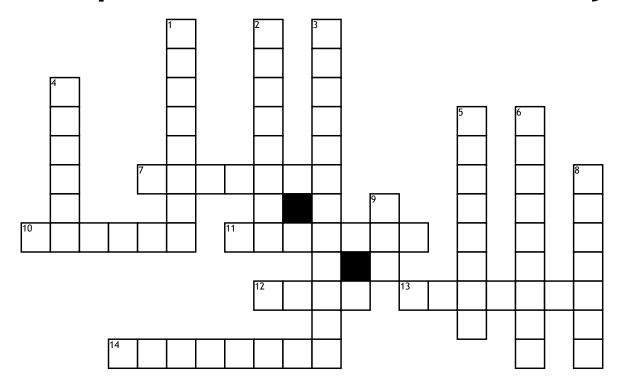
Name:	Date:
-------	-------

## Chapter 4 Earth chemistry



## **Across**

- **7.** a combination of two or more substances that are not chemically combined
- 10. a subatomic particle that has a positive charge and that is located in the nucleus of an atom; the number of protons of the nucleus is the atomic number, which determines the identity of an element
- **11.** a subatomic particle that has no charge and that is located in the nucleus of the atom
- **12.** the smallest unit of an element that maintains the chemical properties of that element

- 13. one of two more atoms that have the same number of protons (atomic number) but different number of neutrons
- **14.** a substance made up of atoms of two or more different elements joined by chemical bonds

## <u>Down</u>

- 1. a homogeneous mixture throughout which two or more substances are uniformly dispersed
- **2.** a group of atoms that are held together by chemical forces
- **3.** a bond formed when atoms share one or more pairs of electrons
- anything that has mass and takes up space

- **5.** a subatomic particle that has a negative charge
- **6.** the attractive forces between oppositely charged ions, which form when electrons are transferred from one atom or molecule to another
- **8.** A substance that cannot be separated or broken down into simpler substances by chemical means
- **9.** an atom or molecule that has gained or lost one or more electrons and has a negative or positive charge