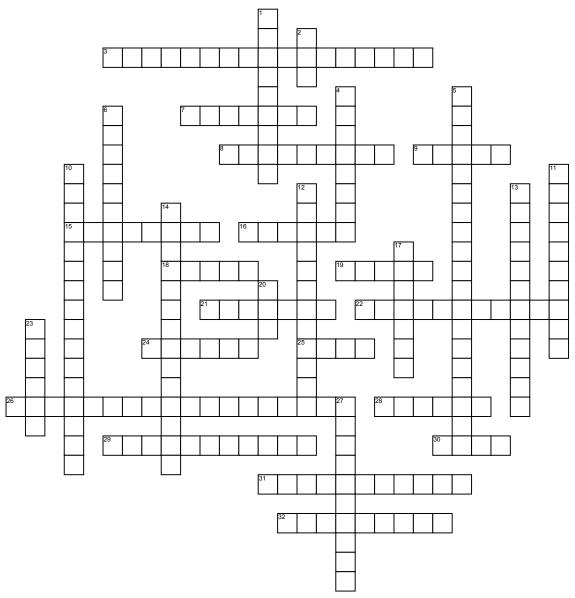
Name:	Date:	Period:

## Chem Part1 Vocab



## Across

- 3. the weighted average of the atomic masses of the naturally occurring isotopes of an element
- 7. general term for any isotope of any element
- **8.** the number of waves that passes a given point in a specific time
- **9.** an element that is a good conductor of heat and electricity
- **15.** an element that is a poor conductor of heat and electricity
- **16.** broad generalization that explains a boy of facts or phenomena
- 18. more than a physical object
- 19. vertical row of elements on periodic table
- **21.** a pure substance made only of one kind of atom
- 22. 14 elements from atomic number 58 to 71
- 24. positive ion

- **25.** amount of a substance that contains sa many particles as there are atoms in exactly 12 gs of carbon-12
- **26.** the emission of a continuous range of frequencies of electromagnetic radiation
- 28. a specific portion of matter in a given region of space that has been selected for study during an experiment
- **29.** Physical and chemical properties of the elements are periodic functions of their atomic numbers
- **30.** smallest particle of an element that retains the chemical properties of that element
- 31. lowest energy state of an atom
- 32. the class we are all in

## <u>Down</u>

- **1.** 14 elements from atomic number 91 to 103
- **2.** a form of energy that exhibits wavelike behavior as it travels through space

- **4.** something that has magnitude, size, or amount
- **5.** Refers to the emission of electrons from a metal when light shines on the metal
- **6.** distance between corresponding points on adjacent waves
- 10. metals with typically metallic properties
- 11. a testable statement or educated guess
- **12.** number of protons in the nucleus of each atom of that element
- **13.** higher potential energy than it is at the ground state
- **14.** 1 amu is exactly 1/12 the mass of carbon-12 atom
- **17.** the minimum quantity of energy that can be lost or gained by an atom
- 20. all the forms of electromagnetic radiation23. horizontal row of elements on periodic
- 23. horizontal row of elements on periodic table
- **27.** number of protons and neutrons in the nucleus of an isotope