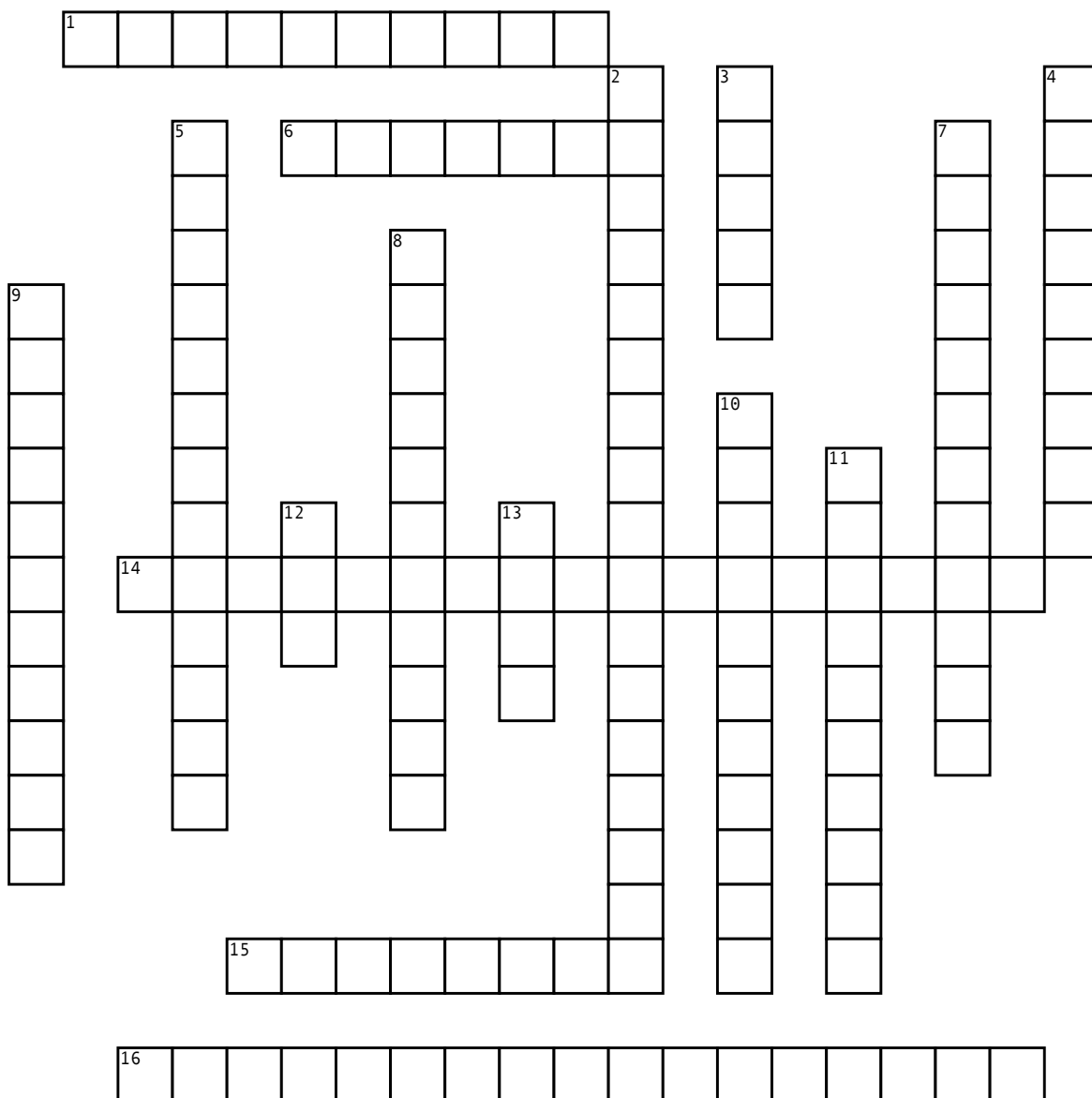


Name: \_\_\_\_\_

Date: \_\_\_\_\_

# Chem Unit 5 Crossword



## Across

1. Finish and balance Double replacement reaction  $\text{AgNO}_3 + \text{NaCl} \rightarrow$   
 6. Finish combustion  $\text{CH}_4 + \text{O}_2 \rightarrow$   
 14. Name the reaction  $\text{Zn} + \text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$   
 15. Something that increases the rate of a chemical reaction  
 16.  $\text{Cl}_2 + \text{NaBr} \rightarrow 2\text{NaCl} + \text{Br}_2$ , Balance the equation

## Down

2. Name the reaction  $\text{NaCl} + \text{H}_2\text{SO}_4 \rightarrow \text{Na}_2\text{SO}_4 + \text{HCl}$

3. If 15 grams of sodium hydroxide are actually produced, what was my percent yield?  
 4. name the reaction  $\text{K} + \text{Cl}_2 \rightarrow \text{KCl}$   
 5. Name the reaction  $\text{H}_2\text{O} \rightarrow \text{H}_2 + \text{O}_2$   
 7. If 25 grams of carbon dioxide gas is produced, how many grams of sodium hydroxide will be produced?  $\text{NaHCO}_3 \rightarrow \text{NaOH} + \text{CO}_2$   
 8. What reaction takes in heat

9. one substance that changes the rate of a chemical change  
 10. Finish the single replacement reaction  $\text{Cu} + \text{AgNO}_3 \rightarrow$  and balance  
 11. Name the reaction  $\text{C}_6\text{H}_{12}\text{O}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$   
 12. Finish synthesis  $\text{Mg} + \text{O}_2 \rightarrow$   
 13. Finish decomposition  $\text{FeS} \rightarrow$