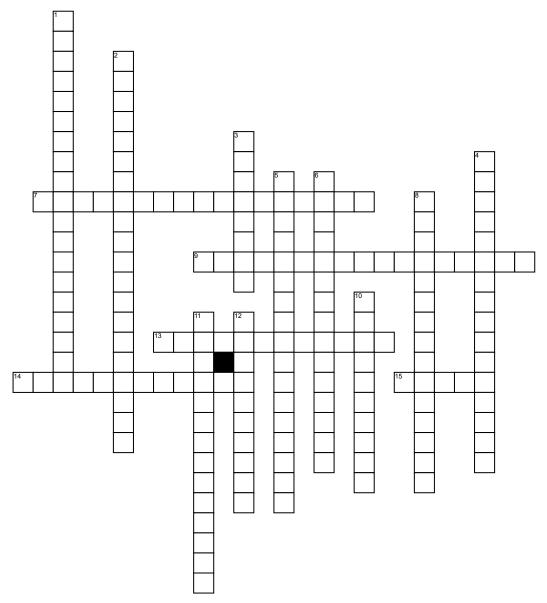
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Chemical Bonding



Across

- **7.** bonding is a type of chemical bond where a pair of electrons is unequally shared between two atoms.
- **9.** atoms share electrons through covalent bonds.
- **13.** a lasting attraction between atoms, ions or molecules that enables the formation of chemical compounds.
- **14.** formed between two ions with opposite charges.
- **15.** a covalent bond between two atoms where the electrons forming the bond are unequally distributed.

Dowr

- 1. electrons that are represented by dots; two dots between atoms represent a covalent bond.
- **2.** two atoms share a pair of electrons with each other.
- **3.** a group of atoms or 1 or 2 atoms or more
- **4.** a formula giving the number of atoms of each of the elements present in one molecule of a specific compound.
- **5.** a formula that shows the arrangement of atoms in the molecule of a compound.

- **6.** A chemical bond formed by the sharing of one or more electrons, especially pairs of electrons, between atoms
- **8.** tells us the number of atoms of each element in a compound.
- **10.** energy required to break apart a mole of molecules into its component atoms
- **11.** diagrams that show the bonding between atoms of a molecule
- **12.** one pair of electrons is shared by two atoms in a molecule