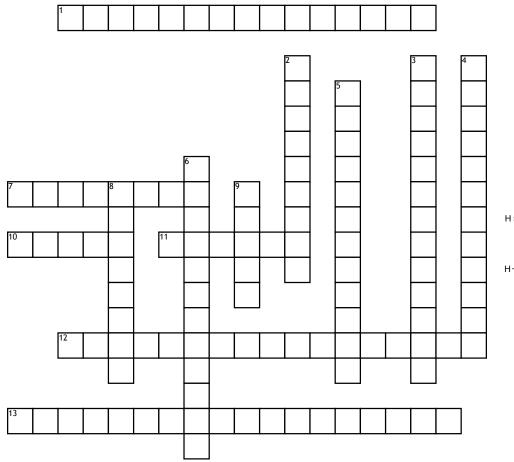
Name: _____ Date: _____

Chemical Bonds



Across

- 1. a notation that shows what elements a compound contains and the ratio of the atoms
- 7. a neutral group of atoms that are joined together by one or more covalent bonds
- **10.** a mixture of two or more elements
- **11.** an ion with a positive charge
- **12.** a covalent bond in which electrons are not shared equally

13. a model of an atom in which each dot represents a valence electron

Down

- **2.** the force that holds cation and anions together
- **3.** a covalently bond group of atoms that has a positive or negative charge and as a unit
- **4.** a chemical bond in which two atoms share a pair of valence electrons

- **5.** the force that holds atoms or ions together as a unit
- **6.** is the attraction between a metal cation and the shared electrons that surround it
- **8.** solids whose particles are arranged in a lattice structure
- **9.** an ion with a negative charge