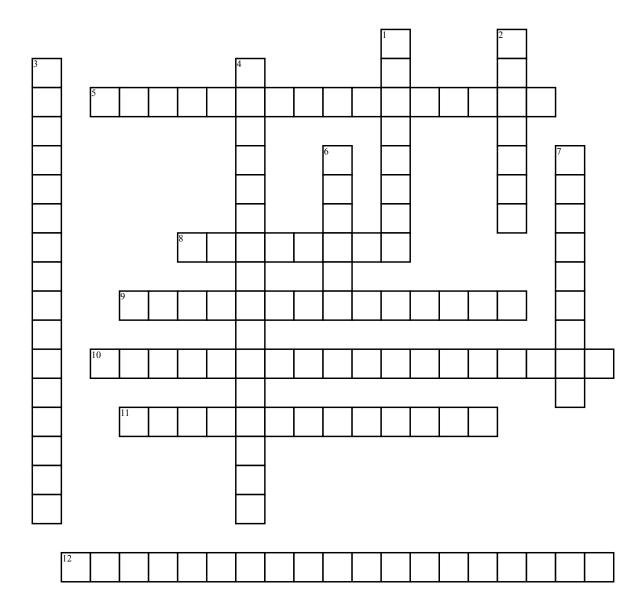
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Chemical Reactions



Across

- **5.** a process that involves rearrangement of the molecular or ionic structure of a substance, as opposed to a change in physical form or a nuclear reaction.
- **8.** substance that speeds up chemical reactions but is not used up itself
- **9.** measure of how fast a chemical reaction occurs
- **10.** chemical reaction in which heat energy is released

- 11. describes how much solute is present in a solution compared to the amount of solvent
- **12.** chemical reaction in which heat energy is absorbed

Down

- **1.** a substance that takes part in and undergoes change during a reaction.
- **2.** is a substance that is formed as the result of a chemical reaction
- **3.** A chemical equation is the symbolic representation of a chemical reaction in the form of symbols and formulae, wherein the reactant entities are given on the left-hand side and the product entities on the right-hand side
- **4.** minimum amount of energy needed to start a chemical reaction
- **6.** isgbcxlwhclqhsvyluxhinqi;w
- 7. substance that slows down a chemical reaction making the formation of a certain product