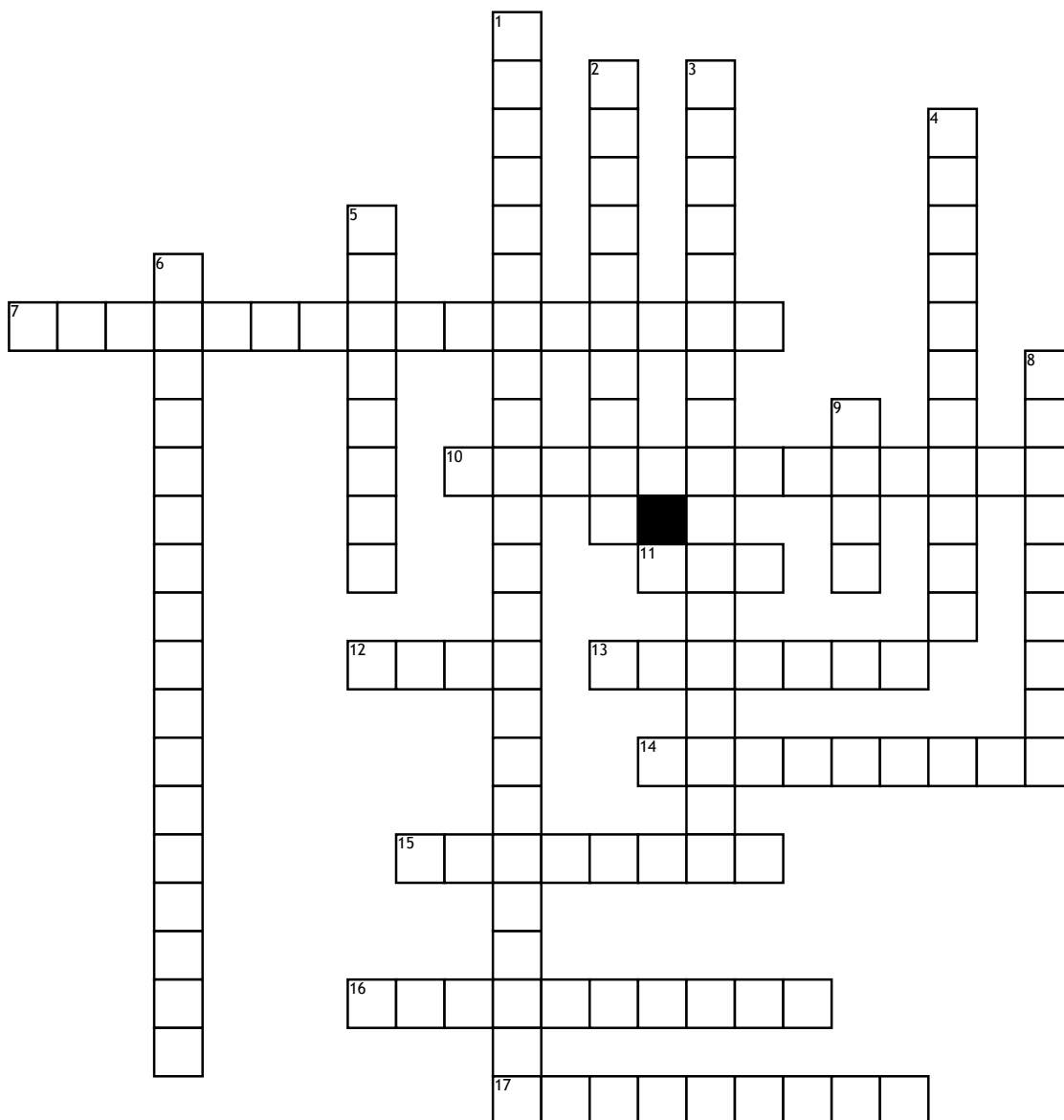


Name: _____

Chemical Reactions



Across

7. This person proposed the Law of Conservation of Mass
10. $AB \rightarrow A+B$ is the generic equation for this type of reaction
11. A _____ could be produced to indicate a chemical reaction has occurred.
12. This is what is formed when the iron in your car reacts with the oxygen in the air.
13. Matter is neither _____ or destroyed during a reaction.
14. These can be found on the left hand side of the arrow in a reaction.

15. These can be found on the right hand side of the arrow in a reaction.
16. Before adding any coefficients, a chemical reaction is _____.
17. This is the small number that goes after an element.

Down

1. This states that the same amount of matter is contained in the reactants and the products
2. Fuel + oxygen makes carbon dioxide and water is the generic equation for this type of reaction

3. $AB+C \rightarrow AC+B$ is the generic equation for this type of reaction
4. This number goes in front of the compound.
5. We can change the coefficients in a chemical reaction to make it _____.
6. $AB+CD \rightarrow AC+BD$ is the generic equation for this type of reaction
8. $A + B \rightarrow AB$ is the generic equation for this type of reaction
9. There are ____ hydrogens in this compound: $5HCl$.