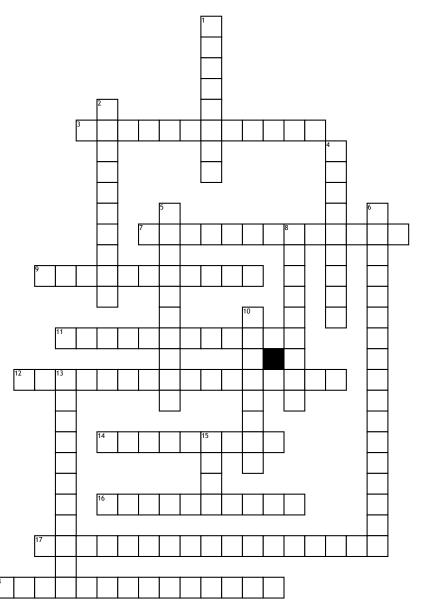
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## Chemical Reactions



## Across

- **3.** the numbers that appear before the formulas
- **7.** a reaction in which a compound breaks down into two or more simpler substances
- **9.** a chemical reaction that absorbs energy from its surroundings
- 11. is the rate at which reactants change into products over time
- **12.** a representation of a chemical reaction in which the reactants and products are expressed as formulas
- **14.** the mass of one mole of a substance is called a
- **16.** a chemical reaction that releases energy to its surroundings

- 17. a reaction in which one element takes the place of another element in a compound
- **18.** is the energy stored in the chemical bonds of a substance

## Down

- 1. the new substances formed as a result of that change are called
- **2.** one in which a substance reacts rapidly with oxygen
- **4.** in a chemical reaction the substances that undergo change are called
- **5.** a reaction in which the conversion of reactants into products

- **6.** one in which two different compounds exchange positive ions and form two new compounds
- **8.** a reaction in which two or more substances react to form a single substance is called a
- **10.** a substance that affects the reaction rate without being used up in the reaction
- **13.** a state in which the forward and reverse paths of a change take place at the same rate
- **15.** is an amount of a substance that contains approximately 6.02 \* 10 particles of that substance