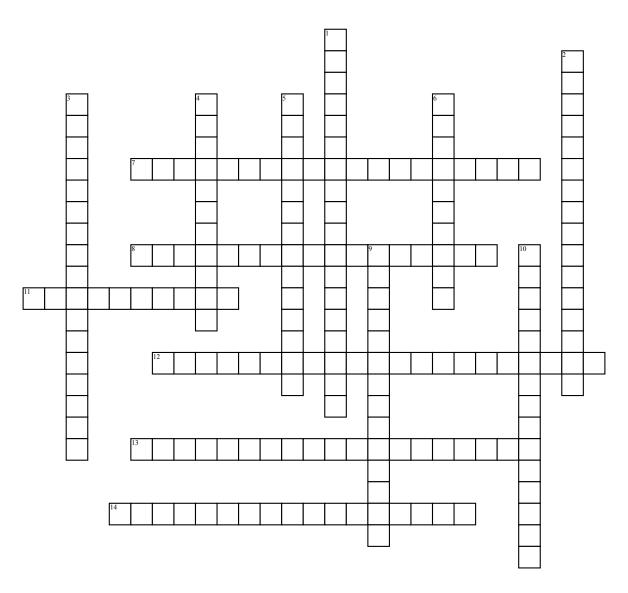
Chemical Reactions



Across

- 7. reaction that produces a precipitate
- **8.** 2 or more substances combine to create a more complex pruduct
- **11.** property of a substance to dissolve in a solute
- **12.** a single compound breaks down into 2 or more elements or new compounds
- **13.** one atom or compound "steals" electrons from another atom or compound

14. one element replaces another element in a compound

Down

- **1.** a reaction with oxygen (aka burning)
- **2.** an equation showing the charges and elements of the precipitate compound
- **3.** 2 compounds react so that cation and anion pairs switch places
- **4.** insoluble ionic solid product of a chemical reaction

- **5.** when an acid and a base react quantitatively
- **6.** tendency of a substance to undergo chemical changes
- **9.** series of metals arranged by reactivity from highest to lowest
- **10.** number that represents the number of electrons lost or gained in a redox reaction