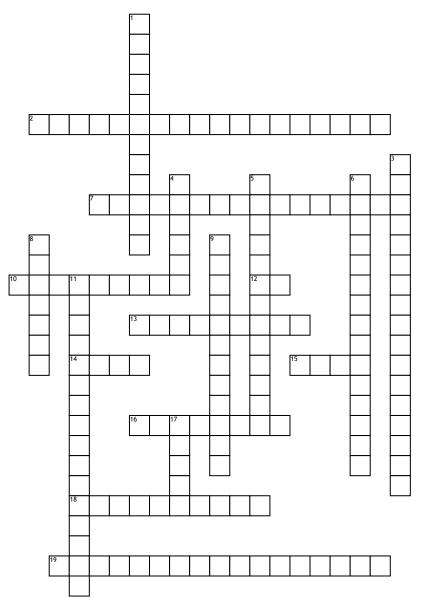
## Chemical Reactions Vocab.



## <u>Across</u>

percent by mass of each element in a compound
 using chemical formulas and other symbols
 mass of one mole of a pure substance
 hydrogen ion concentration
 combining parts into a whole
 compound that produces hydroxide ions
 used to measure the amount of a substance

**16.** is the start of a chemical reaction

**18.** rapid reaction between oxygen and fuel that results in fire

**19.** two elemennt compounds trade places

## <u>Down</u>

1. ratio of the actual yield to the theoretical yield expressed as a percent

**3.** 1 element replaces 1 other element

**4.** amount of product obtained in a chemical reaction

5. breaks down compounds into simpler products6. amount of product that can be formed in the reaction

8. substance produced in a chemical reaction
9. Study of quantitative relationships that can be derived from chemical formulas and equations
11. acid and base mix

together, neutralize one another's properties **17.** compound that forms H+ ions