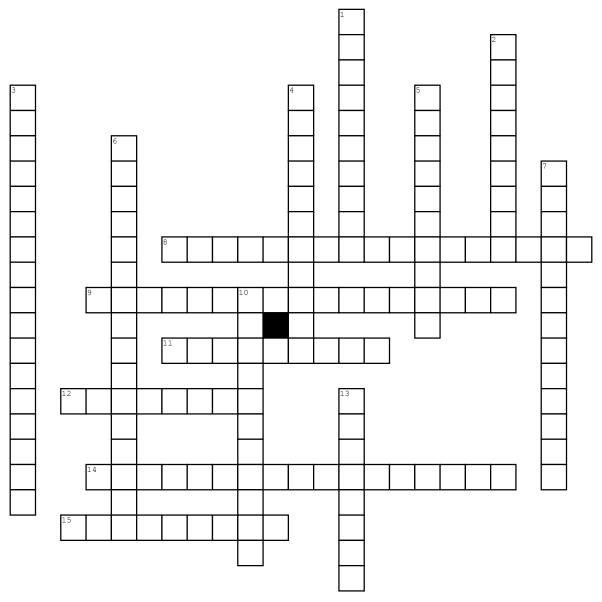
Chemical Reactions & Law Conservation



Across

- 8. the most fundamental law in chemistry
- 9. What shows the Law of Conservation of Mass?
- 11. a chemical change in which a substance combines with oxygen, as when iron oxidizes, forming rust
- 12. the starting materials in a chemical reaction
- 14. occurs when the atoms of one element replace the atoms of another element in 4. tells how many atoms the compound
- 15. Combine to form a single product

<u>Down</u>

- 1. oxygen combines with a substance and releases energy in the form of heat and light
- 2. a number in a chemical formula that tells the number of atoms in a molecule or the ratio of elements in a compound
- 3. occur when ions exchange between two compounds
- molecules or formula units take part in a reaction
- 5. Releases energy

- 6. a process during which chemical bonds between atoms are broken down and new ones are formed
- 7. A reaction in which a single compound breaks down to form two or more simpler substances
- 10. Absorbs energy
- 13. a substance that increases the rate of a chemical reaction without itself undergoing any permanent chemical change.