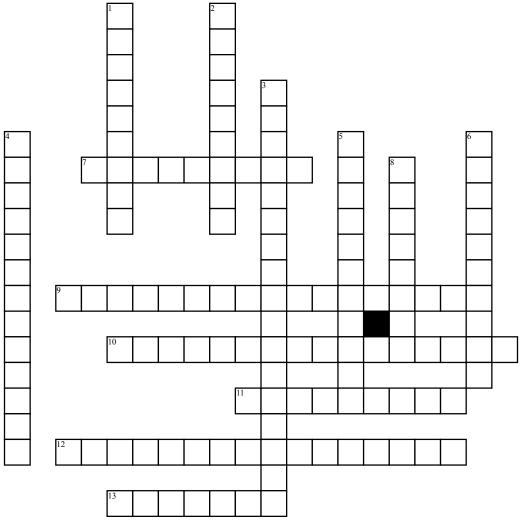
Chemical reaction



Across

- **7.** The sign between the reactants and the products
- **9.** The positive and negative ions in two compound switch places
- **10.** A chemical change in which two or more substances combine to form a new substance
- **11.** Left side of a chemical formula

- **12.** A representation of a chemical reaction using symbols of the elements
- **13.** Right side of a chemical formula

Down

- **1.** Tells you how many atoms of an element
- 2. Compound made from simpler materials
- **3.** One element that starts out by itself replaces another element in a compound
- **4.** A compound broken down into simpler compound
- **5.** Tells you how many molecules or atoms in a reaction
- **6.** A compound containing carbor, hydrogen, and some times oxygen companies with oxygen gas to produce carbon dioxide and water
- **8.** Appears in nature in pair

Word Bank

Yield sign Single replacement Combustion Synthesis

Double replacement Reactants Decomposition Chemical equation

Diatomic Chemical reaction Coefficient Subscript

Product