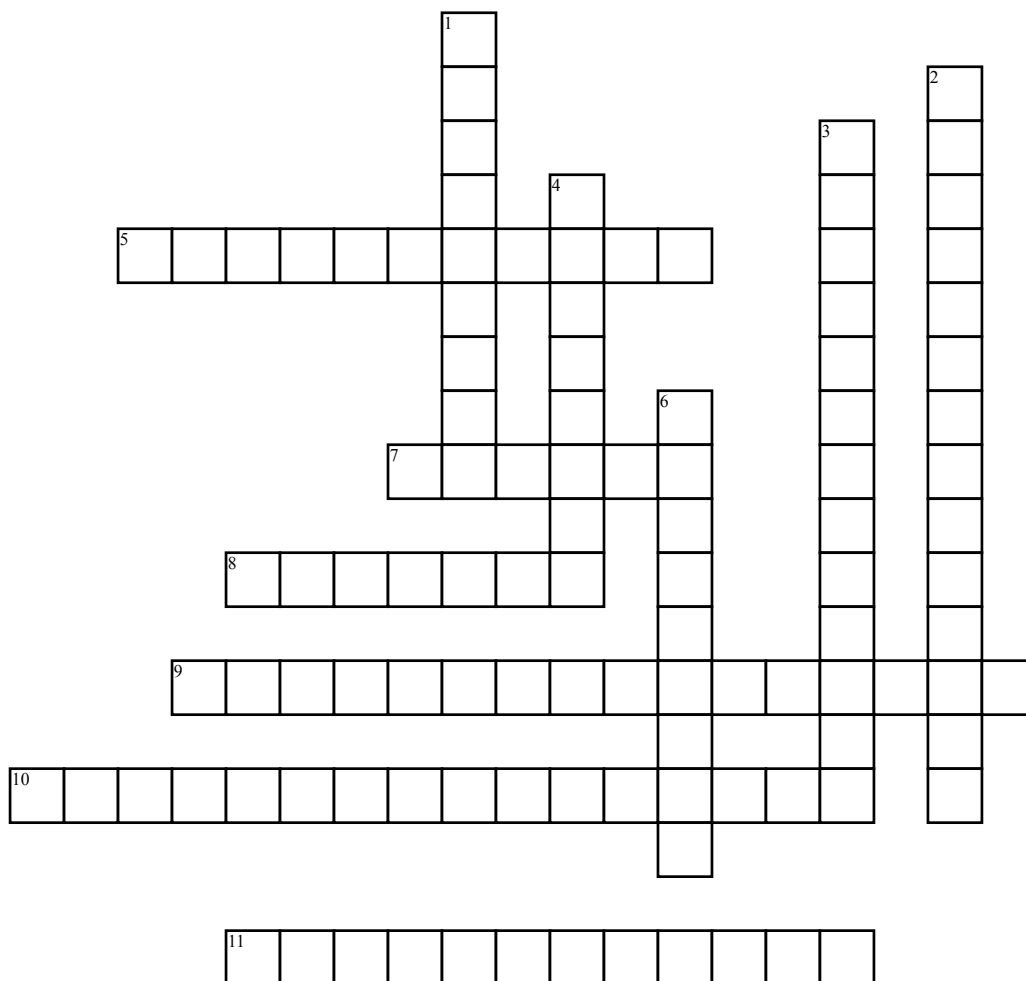


Name: _____

Chemical reactions JL



Across

- 5. A reaction that happens when two liquids react to become solid.
- 7. A biological catalyst found in cells of organisms
- 8. The chemicals you get after the reaction has taken place
- 9. Another term for the speed of a reaction
- 10. When chemicals react together to form a new product.
- 11. When we decrease the size of this, the surface area increases and collision is more likely

Down

- 1. The reacting particles must do this for a reaction to take place
- 2. Similar to a chemical reaction, however a new product is not formed, and you can reverse what you have done.
- 3. Increasing this in a reactant increases the number of particles therefore increases the chance of collision.
- 4. This will increase the rate of a reaction product
- 6. The chemicals you start off with before a reaction

Word Bank

Chemical Reaction	Rate (of reaction)	Concentration	Product
Collision	Particle Size	Catalyst	Enzyme
Precipitate	Reactants	Physical Change	