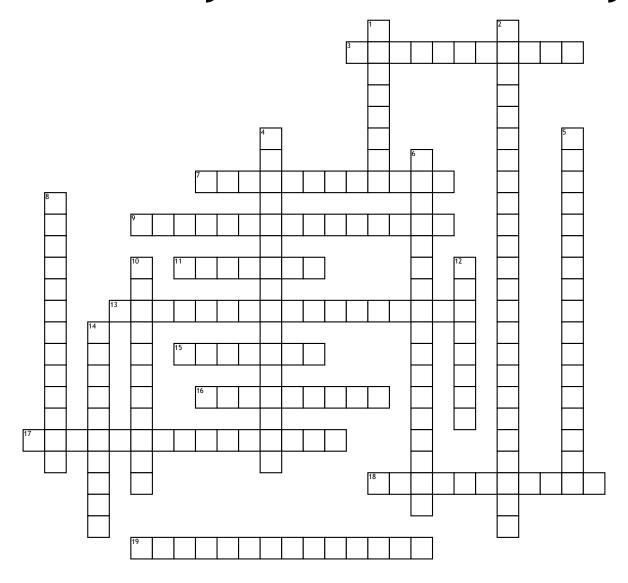
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Chemistry Ch. 8 Vocabulary



Across

- **3.** A molecule that contains three different elements hydrogen, oxygen, and another nonmetal
- **7.** A system of naming that follows a standardized set of rules
- **9.** The number of electrons that an atom in a compound must gain or lose to return to its neutral state
- **11.** A compound that holds a characteristic number of water molecules within its crystalline structure
- 13. A chemical reaction of the general form AY + BZ \rightarrow AZ + BY, in which the cation of one compound combines with the anion of another compound, or vice versa
- **15.** An element or compound that is the result of a chemical reaction and is written to the right of the arrow in a chemical equation

- **16.** A chemical reaction of the general form $A+B\to AB$, in which two or more reactants combine into a single product
- **17.** An ionic compound that contains polyatomic ions
- **18.** A solid that settles out of a solution as the result of physical or chemical changes
- **19.** A compound consisting of two different nonmetal atoms

<u>Down</u>

- 1. An element or compound that undergoes a chemical change and is written to the left of the arrow in a chemical equation
- **2.** A chemical equation in which coefficients are arranged to show the conservation of mass in a reaction
- **4.** An expression that represents the reactants and products in a chemical reaction by using chemical formulas, symbols, and numerical coefficients

- **5.** A chemical reaction of the general form $A + BZ \rightarrow B + AZ$, in which a reactive element replaces a less reactive element in a compound
- **6.** Nomenclature that indicates how many atoms of each element are in a binary covalent compound or how many water molecules are in a hydrate
- 8. A chemical reaction of the general form AB \to A + B, in which a reactant breaks down into two or more products
- **10.** A convention used to show the different oxidation states of the same metal ion in an ionic compound
- **12.** A substance that changes a reaction rate without being permanently changed in the process
- **14.** A molecule that consists of hydrogen and one other nonmetal that are dissolved in water