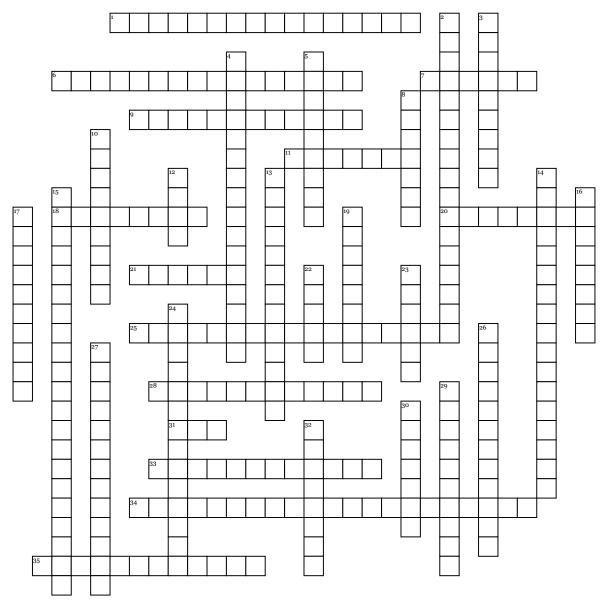
Name:	Date:	Period:

Chemistry Vocabulary Quarter 1



Across

- 1. Number of electrons in the most outer shell
- **6.** Two atoms share a pair of electrons with each other
- 7. Positive charged ion
- 9. They are the metals found in Group 1
- **11.** Composed of atoms that have the same atomic number
- 18. Negative electric charge
- 20. They only need 1 more electron to become full
- 21. Positive electric charge
- **25.** An element is the sum of the masses of its isotopes, each multiplied by its natural abundance
- 28. Sharing of electron pairs between atoms
- **31.** An electrically charged atom or group of atoms formed by the loss or gain of one or more electrons
- **33.** The number of protons in the nucleus of an

- **34.** Distribution of electrons of an atom or molecule in atomic or molecular orbitals
- **35.** Electronegativities based on bond-energy calculations for different elements joined by covalent bonds

Down

- 2. Metals found in Group 2
- $\boldsymbol{3.}$ When one atom gives up one or more electrons to another atom
- 4. Energy required to remove an electron
- **5.** Element whose properties are intermediate between those of metals and solid nonmetals
- 8. Same mass as a proton but no electric charge
- **10.** Chemical element that lacks the characteristics of a metal and that is able to form anions, acidic oxides, acids, and stable compounds with bufferon.
- 12. Basic unit of a chemical matter
- **13.** Pair of electrons are unequally shared between two atoms
- 14. Energy to gain an electron

- **15.** Shows the bonding between atoms of molecules and their lone pairs
- **16.** Equal numbers of protons but different number of neutrons
- 17. Total number of protons and neutrons
- 19. Consisting of two atoms
- 22. Negativity charged ion
- **23.** Chemical element that is an effective conductor of electricity and heat
- **24.** Includes electronegativity, ionization energy, and atomic radius
- **26.** Chemical element measure based on size of atom
- **27.** Table of the chemical elements arranged in order of atomic number
- **29.** They cannot gain anymore electrons
- 30. Positive charge center of an atom
- **32.** Substance formed when two or more chemical elements are chemically bonded together