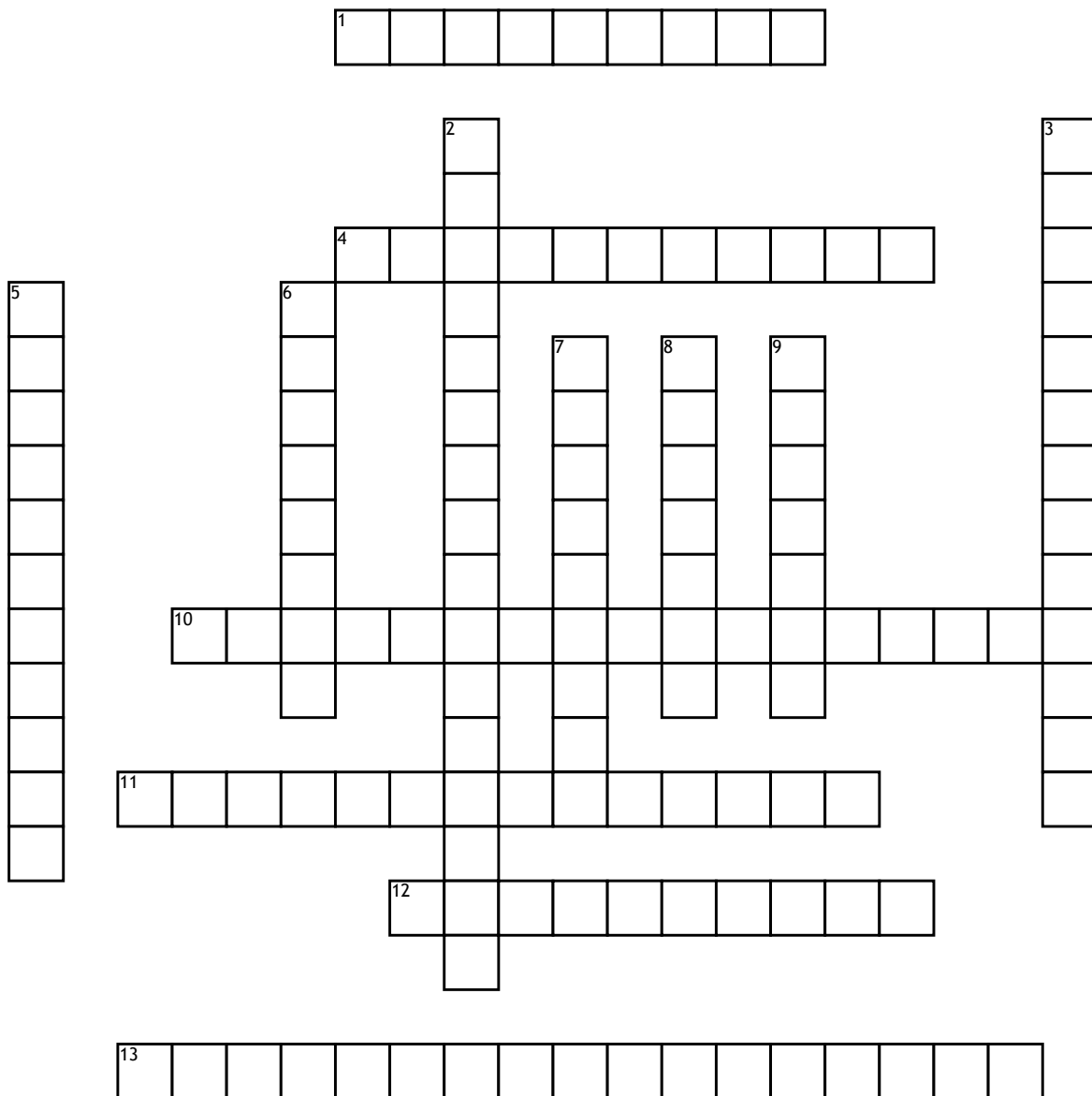


Name: _____

Date: _____

Chemistry



Across

1. All carbonates and phosphates except group 1 and ammonium are _____

4. A small whole number that appears in front of a formula in a chemical equation is called a _____

10. $\text{Ag} + \text{CuSO}_4$ is what type of replacement

11. $\text{F}_2/\text{Cl}_2/\text{Br}_2/\text{I}_2$ are _____ of halogens

12. Hydrocarbon only consists in what type of reaction

13. $2\text{NaI} + \text{CaCl}_2$ is what type of replacement

Down

2. When the substance changes it's called a _____

3. $\text{O}_3 \rightarrow \text{O} + \text{O}_2$ is what type of reaction

5. A solid that is produced as a result of a chemical reaction

6. A substance that reacts in a chemical change is called a _____

7. $\text{O}_2 + 2\text{H}_2$ is what type of solution

8. All nitrates are _____

9. A substance that is formed by a chemical change is called a _____