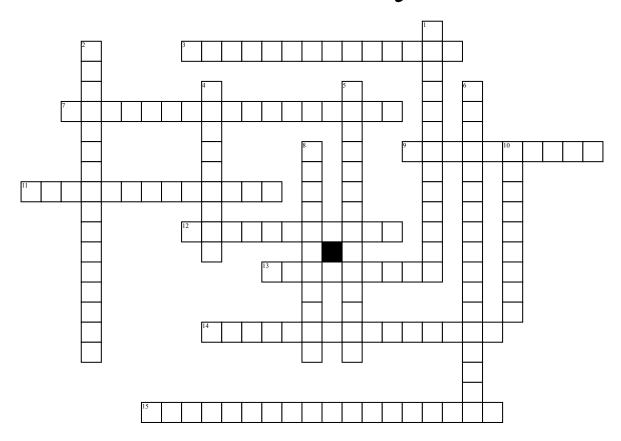
## Chemistry



## **Across**

- **3.** A guide for predicting metal displacement reaction
- 7. Element reacts with a compound and takes the place of another element
- **9.** Reaction which something reacts with oxygen and water is a product
- 11. Reaction where two soluble salts produce an insoluble salt
- **12.** Property of a solid liquid or gas

- **13.** A reaction where a compound steals electrons from another compound
- **14.** A number an element has that represents lost electrons
- **15.** The likelihood of a substance to release ene rgy and to undergo chemical reaction

## Down

- **1.** A reaction of breaking down chemical compounds into simpler compounds
- **2.** An equation that shows only the soluble elements

- **4.** A chemical reaction which the oxidation states change
- **5.** Reaction which and acid and a base react
- **6.** Reaction where negative ions and positive ions switch places to form a new compound
- **8.** A soluble substance from a solution
- **10.** Most common type of chemical reaction