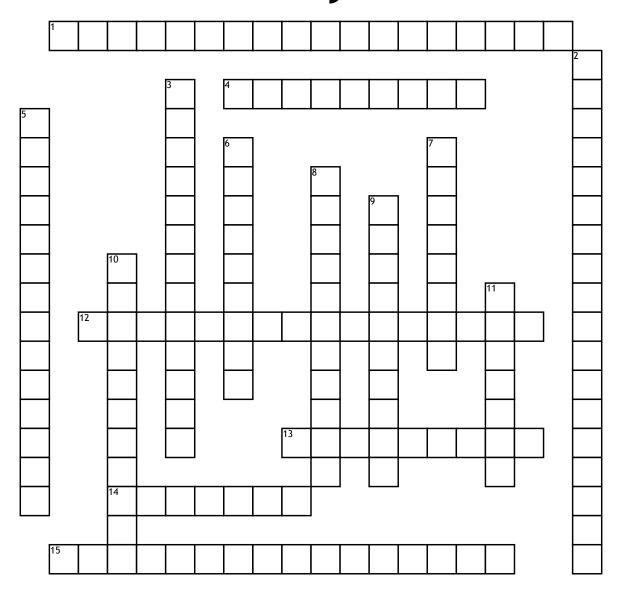
Name:	Date:	Period:
-------	-------	---------

## Chemistry vocab



## **Across**

- 1. where an element reacts with a compound and takes the place of another element in that compound
- **4.** the production of chemical compounds by reaction from simpler materials
- **12.** a chemical equation for a reaction which lists only those species participating in the reaction
- **13.** substance is a substance that will not dissolve in a solvent even after mixing
- **14.** a solution in which the solvent is water

**15.** a process that involves rearrangement of the molecular or ionic structure of a substance,

## **Down**

- 2. where two compounds react, and the positive ions and the negative ions of the two reactants switch places,
- **3.** the separation of a single chemical compound into its two or more elemental parts or to simpler compounds.
- **5.** an list of metals ranked in order of decreasing reactivity to displace hydrogen gas from water and acid solutions

- **6.** a substance that takes part in and undergoes change during a reaction
- **7.** a substance that is formed as the result of a chemical reaction
- **8.** a number in front of a chemical
- **9.** is any process in which a substance combines with oxygen to produce heat and light
- **10.** an insoluble solid that emerges from a liquid solution.
- 11. property referring to the ability for a given substance