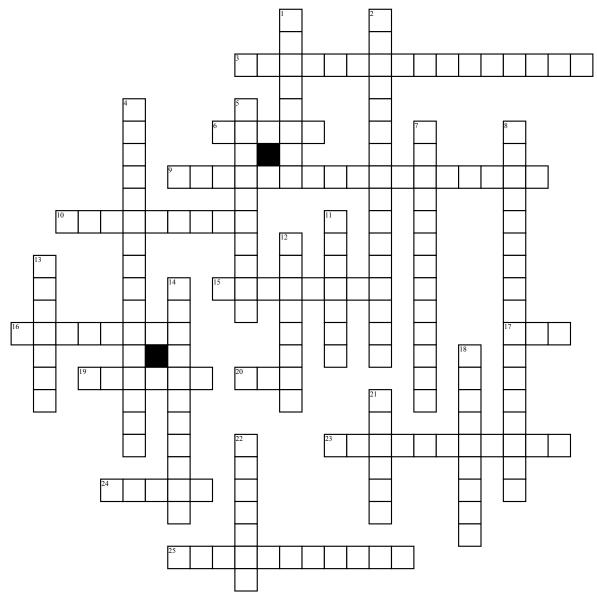
Name:	Date:	Period:

Crossword: Chemistry H



Across

- **3.** having the same number and types of atoms on reactant and product sides of an equation
- **6.** (of a chemical bond) formed by the electrostatic attraction of oppositely charged ions. Often contrasted with covalent.
- **9.** 2 aqueous ionic compounds combine to form a new aqueous ionic compound and a new aqueous solution
- **10.** 2 or more reactants combine to form one product
- **15.** relating to or denoting chemical bonds formed by the sharing of electrons between atoms. Often contrasted with ionic.
- **16.** is composed of two or more separate elements; a mixture.
- **17.** matter cannot be created or destroyed (answer is abbreviated)

- 19. a positively charged ion
- **20.** abbreviation for "reaction"
- **23.** the only thing you can change in a chemical equation to balance
- **24.** an ionic species having a negative charge
- **25.** insoluble solid that emerges from a liquid solution

Down

- 1. measured value that is "with respect for"
- 2. a process that involves rearrangement of the molecular or ionic structure of a substance 4. an unbalanced equation, whereby the
- chemical names in a word equation are substituted for chemical formulae **5.** a reaction in which oxygen reacts with
- **5.** a reaction in which oxygen reacts with a hydrocarbon

- **7.** one reactant breaks down into 2 or more products
- **8.** a reaction where the single element replaces part of a compound
- **11.** a compound is dissolved in water, a water solution
- **12.** beiginning substance in a chemical equation
- 13. any of the six nonmetallic elements that constitute Group 17 of the periodic table
- **14.** a compound of both hydrogen and carbon
- **18.** a distinguishing symbol (such as a letter or numeral) written immediately below or below and to the right or left of another character.
- **21.** the way we read the arrow in a chemical equation
- 22. a substance produced in a chemical reaction