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## Gas Laws Crossword Puzzle



## Across

2. If something is constant or not mentioned, then $\qquad$ use it to solve the problem
3. Atm and mmHg are units for $\qquad$
4. Pressure unit that you must be in for the Ideal Gas Law
5. Gas Law using Volume and Moles
6. Type of Relationship for Pressure and Temperature
7. Milliliters must be converted to $\qquad$
8. If moles and pressure are constant, then as Temperature increases, Volume $\qquad$ Down
9. Type of relationship for Pressure and Volume
10. Gas Law using Pressure and Volume

## Word Bank

Avogadro's Law Increases
Inverse

Ideal Gas Law
Boyle's Law Liters
don't
Charles' Law
Kelvin

Pressure Direct decreases

Temperature atm
4. Gas Law using Temperatue and Volume
5. If moles and temperature are constant, then as Pressure increases, Volume $\qquad$
6. Kelvin is a unit for $\qquad$
7. Gas Law with ONE set of Conditions
9. Temperatures must be converted to $\qquad$

