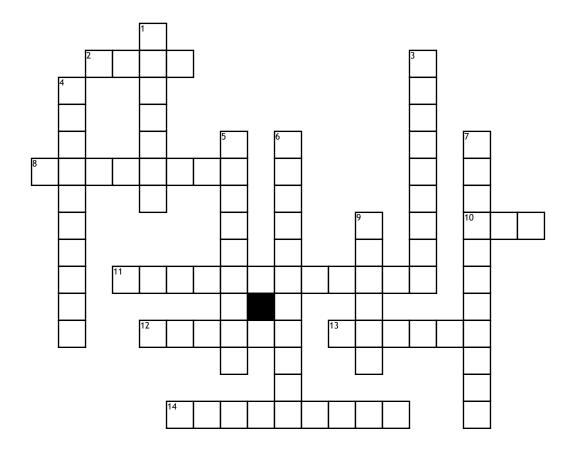
Gas Laws Crossword Puzzle



<u>Across</u>

- **2.** If something is constant or not mentioned, then ____ use it to solve the problem
- **8.** Atm and mmHg are units for ____
- **10.** Pressure unit that you must be in for the Ideal Gas Law
- **11.** Gas Law using Volume and Moles

Liters

- **12.** Type of Relationship for Pressure and Temperature
- **13.** Milliliters must be converted to
- **14.** If moles and pressure are constant, then as Temperature increases, Volume ____

Down

- **1.** Type of relationship for Pressure and Volume
- **3.** Gas Law using Pressure and Volume

- **4.** Gas Law using Temperatue and Volume
- **5.** If moles and temperature are constant, then as Pressure increases, Volume ____
- **6.** Kelvin is a unit for ____
- **7.** Gas Law with ONE set of Conditions
- **9.** Temperatures must be converted to ____

decreases

Word Bank

Inverse

Avogadro's Law Ideal Gas Law don't Pressure Temperature Increases Boyle's Law Charles' Law Direct atm

Kelvin