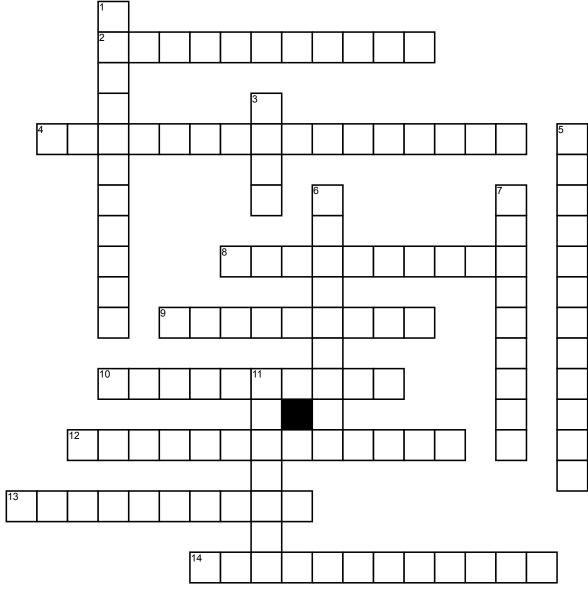
Name:	Date:
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## Heat/Thermal Energy



## **Across**

- 2. When temperatures eventually even out
- **4.** Expansion of matter when it is heated
- 8. Transfer of heat by touch
- **9.** A material in which heat/energy can not move readily
- **10.** Movement of energy through a fluid
- **12.** Total potential and kinetic energy of the particles in an object

- 13. 32 degrees water freezes
- **14.** Temperature at which no other energy can be removed from matter

## Down

- **1.** Measure of average kinetic energy of the particles of matter
- **3.** Thermal energy that is transferred from matter at a high temperature to matter at a lower temperature
- **5.** Amount of energy required to raise the temperature of 1 kilogram of a material one kelvin
- **6.** Transfer of energy through empty space
- 7. A material in which heat/energy can move readily
- **11.** How temperature is measured in metric system