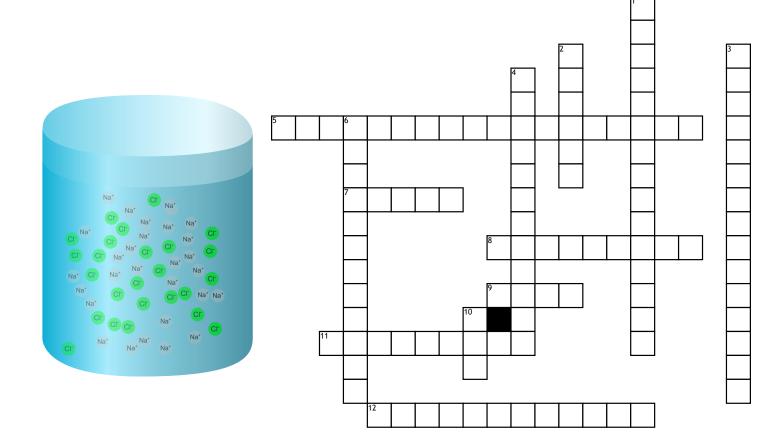
Name: \_\_\_\_\_ Date: \_\_\_\_\_

## **Ionic Bonding**



## **Across**

- **5.** displays valence electrons in an element
- 7. ion with negative charge
- **8.** primary reason why elements form compounds
- **9.** force that holds the atoms together in a substance
- 11. transfer of electrons

12. force that holds atoms or ions together as a unit

## Down

- 1. tells you how many atoms of each element combine
- **2.** ion with positive charge
- **3.** notion that shows what elements a compound contains and the ratio of atoms or ion

- 4. attraction between a metal cation and the shared electrons that surround it
- **6.** sharing of electrons
- 10. atom with a + orcharge