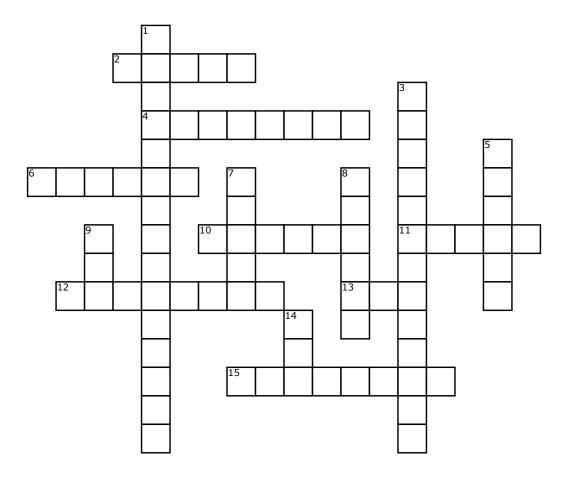
## Ionic VS Covalent Bonds



## **Across**

- **2.** A covalent bonds' state of matter is a \_\_\_\_\_.
- **4.** Covalent bond involves the sharing of \_\_\_\_\_ pairs.
- **6.** An ionic bonds' state of matter is a
- **10.** These bonds are not until they bond.
- **11.** A bond that involves the attraction of oppositely charged ions

- 12. Covalent bonds' create a
- **13.** When naming a bond they contain this ending.
- **15.** A covalent bond requires 2 of these to become stable.

## Down

- 1. An ionic bond has
- 3. A covalent bond has a
- **5.** A metal

- **7.** An ionic bond requires a and a nonmetal to be stable
- **8.** What literary term does a covalent bond have at the beginning of an element?
- **9.** These both connect \_\_\_\_ atoms.
- **14.** An ionic bond creates an when formed.

## **Word Bank**

ide two metal stable ion prefix no definite shape molecule electron cation ionic liquid

nonmetal definite shape

solid