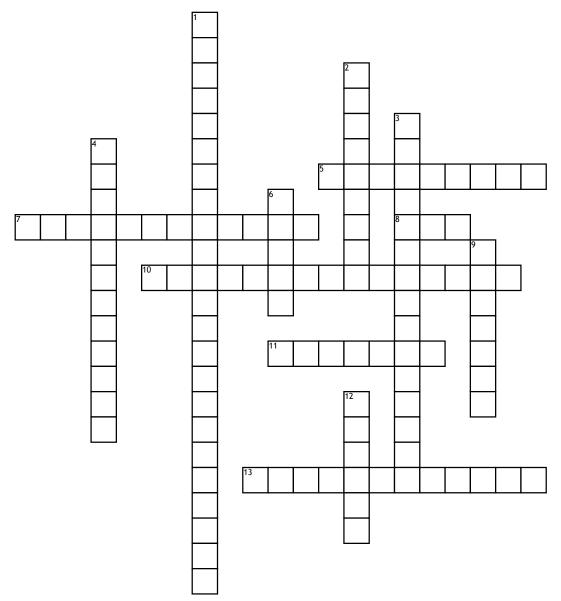
Name:	Date:

Ionic and Covalent Bonding



Across

- **5.** primary reason why elements form compounds
- 7. sharing of electrons
- **8.** a chemical species that has a + or charge
- 10. a way to display the valence electrons in a covalent bond
- 11. this element will harm Clark Kent

13. force that holds atoms or ions together as a unit

<u>Down</u>

- 1. force that holds ions together in an ionic bond
- 2. transfer of electrons
- **3.** tells you how many atoms of each element are in a compound
- 4. attraction between a metal cation and the electrons that surround it6. ion with a negative
- **6.** ion with a negative charge
- **9.** this element will form an ion with a +2 charge in an ionic bond
- **12.** ion with a positive charge