## Isotope Crossword Puzzle

## <u>Across</u>

**4.** Two or more types of atoms that contain the same number of protons but different numbers of neutrons.

**5.** The time taken for the radioactivity of a specified isotope to fall to half of its original value

8. A particle with a positive charge in the nucleus of an atom
14. Photons that carry energy in the form of electromagnetic waves

15. A particle with a neutral charge in the nucleus of an atom
16. When an atom has an uneven number of protons and neutrons it is...

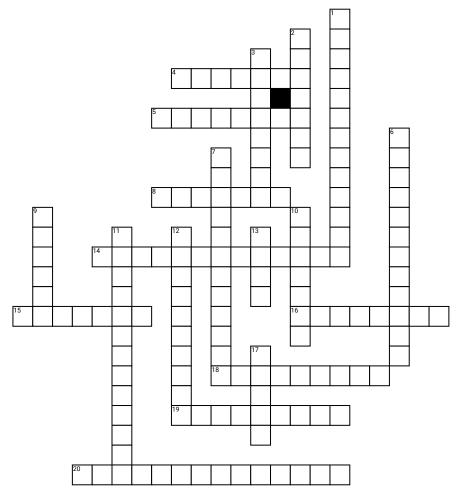
**18.** A particle with a negative charge

**19.** The emission or transmission of energy in the form of waves or particles through space or through a material medium

**20.** Occurs when an atom has more protons than electrons

## <u>Down</u>

**1.** Helium-4 nucleus, a particle consisting of two protons and two neutrons bound together



**2.** A substance that cannot be broken down into any other substance

**3.** The antimatter counterpart of an electron β-

6. Number of protons

7. A high-energy electrons emitted by a radioactive decay of an atomic nucleus

**9.** When an atom has a net electrical charge of 0 it element into a different element, usually with

10. The positively charged central core of atom, consisting of protons and neutrons
11. Elements that undergo radioactive decay
12. Number of protons + number of neutrons
13. The smallest unit of ordinary matter that forms a chemical element
17. The change of an element into a different element, usually with some other particles and energy emitted