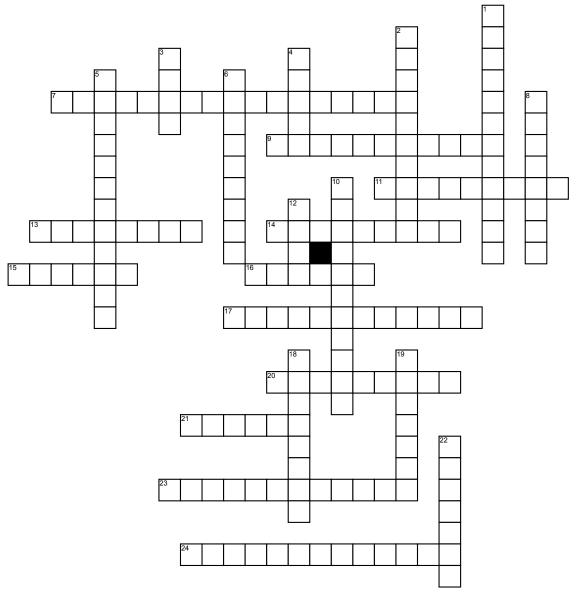
LSA - Aromatic Chemistry



Across

- 7. is the tendency measured by the Pauling scale of an atom to attract the pair of electrons in a bond
- **9.** In an acyclic or cyclic system what is the overlap of more than two 2p orbitals
- **11.** is a reaction in which a compound has a nitro group added to it by a substitution reaction
- **13.** A substance which increases the rate of reaction but remains chemically unchanged at the end of the reaction
- **14.** The name given to the movement of electrons when bonding
- **15.** A three or six membered compound that connects to form a ring
- **16.** All atoms have to be sp2 hybridised
- **17.** A reaction where one functional group of a chemical compound is replaced by another functional group

- **20.** these are dynamic isomers of a compound where the protons and electrons are in different positions from the original compound
- **21.** a rule to determine whether a planar ring molecule would have aromatic properties
- **23.** A reactant which accepts an electron pair which is positive or slightly positive
- **24.** When different groups on the benzene ring influence the position of the subsequent reactions

Down

- 1. When a Chloride group is added within a reaction
- 2. The reaction when an Alkyl group is removed from one molecule to another
- **3.** a molecule with substituents at the 1 and 4 positions on an aromatic compound
- **4.** a molecule with substituents at the 1 and 2 positions on an aromatic compound

- **5.** The type of group that Meta substituents are when reacting
- **6.** Nomenclature name given to a 5-membered heterocyclic ring with 4 N groups within it
- **8.** The compound removed in the Clemmensen reduction reaction
- **10.** Planar molecule where pi bonds are in resonance which cause an increase in stabilization because of the delocalisation of the electrons
- **12.** a molecule with substituents at the 1 and 3 positions on an aromatic compound
- **18.** The reagent used when Thiophene forms 2, 5-dimethyl thiophene
- **19.** A chemical compound formed of six carbon atoms joined in a planar ring with one Hydrogen atom attached to each
- **22.** Nomenclature name given to a 5-membered heterocyclic ring with 1 O and N group within it