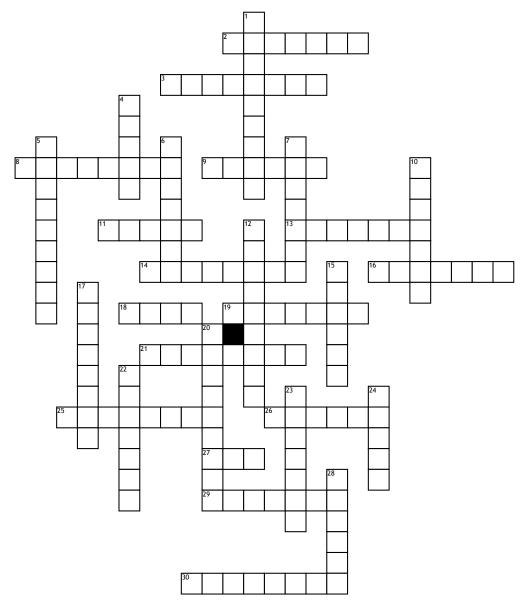
Name:	Date:
-------	-------

## MATTER AND MATERIALS



## **Across**

- 2. The electron that can be lost, gained, shared or transferred
- **3.** This is a bond between non metals only
- **8.** This element has 9 electrons when neutral
- 9. This noble gas has only one period
- 11. This tells us how many electrons are in the outermost orbital.
- 13. A pure substance
- **14.** A non metal found in group one.
- **16.** A metalloid that is found with 4 valence electrons
- **18.** Au
- **19.** This is a subatomic particle that has no charge
- 21. A Halogen with 17 protons
- 25. A pleasant sound when tapped

- **26.** This element can be found in period 6 Group 11
- **27.** An atom with a charge
- **29.** Different substances put together without a bond
- **30.** 2 or more elements chemically bonded in a fixed ratio

## Down

- **1.** Can be bent without breaking or cracking
- **4.** This is a bond between a non metal and a metal
- **5.** This element has 13 electrons when neutral.
- **6.** The only metal that is a liquid at room temperature.
- **7.** the collective name for the subatomic particles in the nucleus

- **10.** This is when something can be pulled into long wires
- 12. This element is in period 3 Group 2
- **15.** A subatomic particle that has a positive charge
- 17. The subatomic particle with a negative charge
- 20. This element has 19 protons
- **22.** A liquid Halogen
- 23. This element has 3 protons
- 24. A noble gas in period 3
- **28.** This tells us how many orbitals in an atom