Matching Type

1. Proposed by Erwin Schrödinger	A. Nuclear Model
2. Matter is neither created nor destroyed. Chemical reaction is just a rearrangement of atoms and molecules	B. Law of Multiple Proportions
3. Proposed by Niels Bohr	C. Billiard Ball Model
4. Suggests that atoms are mostly empty space	D. Electron Cloud Model
5. Suggests that atoms contain negatively charged electrons which are randomly distributed in a sphere of positive charge	E. Plum Pudding Model
6. Same elements can combine in different proportions to form different compounds.	F. Law of Definite Proportions
7. tiny particles	G. atomos
8. Suggests that atoms are tiny, solid particles that make up all materials in the world	H. atom
9. Compounds combine in a fixed proportion	I. Solar System Model
10. A tiny indivisible particle	J. Law of Conservation of Mass