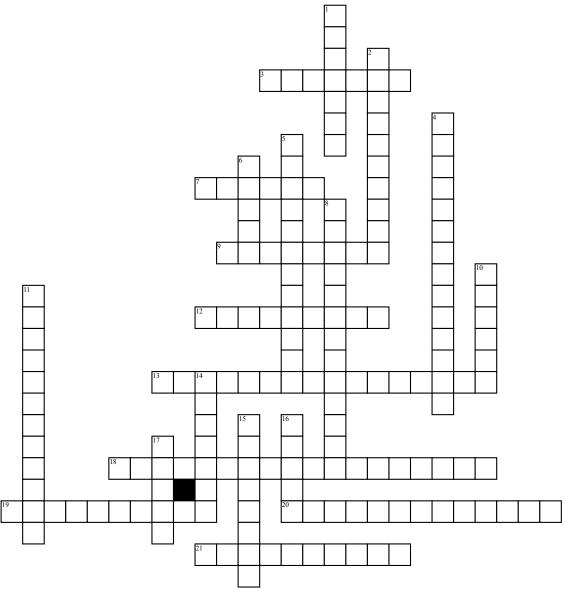
Name:	Date:
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Matter



Across

- **3.** the part of the atom that has no charge
- 7. the part of the atom that is positively charged
- **9.** the part of the atom that moves randomly around the Nucleus
- **12.** The 1913 model of the atom called
- **13.** Who Modeled the 1911 Model of the atom?
- **18.** The newest model of the atom
- **19.** The average mass of all the isotopes of an element is
- **20.** A chart that gives you information about the elements.

21. The ease and speed at which an element reacts

Down

- 1. A material that can be pulled out into a long wire
- **2.** Who came up with the idea of the atom and that it was shaped like a round ball
- **4.** One or two letter representation of an element
- **5.** The number of protons, electrons or _____ are the same number.
- **6.** A mental picture or image that explains something in science

- **8.** A chart that gives you information about the elements.
- **10.** A horizontal row of elements in the periodic table
- **11.** Ability to transfer heat or electricity to another object
- **14.** the part of the atom that protons and neutrons are located
- **15.** Who designed the model of the atom that was made in 1897 and looks like a chocolate chip cookie?
- **16.** Elements in the same vertical column of the periodic table
- **17.** Elements classed by shininess, malleability, ductility, and conductivity