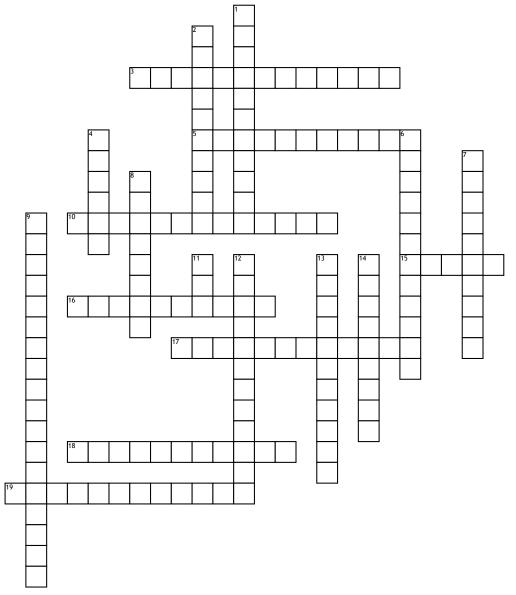
Name:	Date:	

## Matter Project



## <u>Across</u>

- 3. The pressure caused by the collision of particles in a vapor with the walls of a container
- 5. The system absorb energy from it's surroundings
- 10. The energy an object has due to its
- 15. State of matter in which materials have a definite shape and volume
- 16. The direct proportion of the volume of gasto it's tempature if the pressure and the number of particles of the gas are constant
- 17. A tempature of 0 kelvin
- 18. The phase change in which a substance changes from a solid to a gas without changing to a liquid

19. Phase change in which a substance changes from a liquid to a gas

## <u>Down</u>

- 1. The process that changes a substance from a liquid to a gas at tempatures below the substance's boiling point
- 2. The system releases energy to its surroudings
- State of matter in which material has a definite volume but not shape
- 6. The phase change in which a substance changes from a gas to a liquid
- 7. When a gas changes directly into a solid without changing to a liquid
- 8. The result of a force distributed over an area

- **9.** The energy a substance must absorb in order to change from liquid to
- 11. State of matter in which a material has neither a definite shape or volume
- 12. The energy a substance must absorb in order to change from a solid to
- 13. The reversible physical change that occurs when a substance changes from one state of matter to another
- 14. The inverse variation of the volume of a gas with it's pressure if the tempature and the number of particles are constant