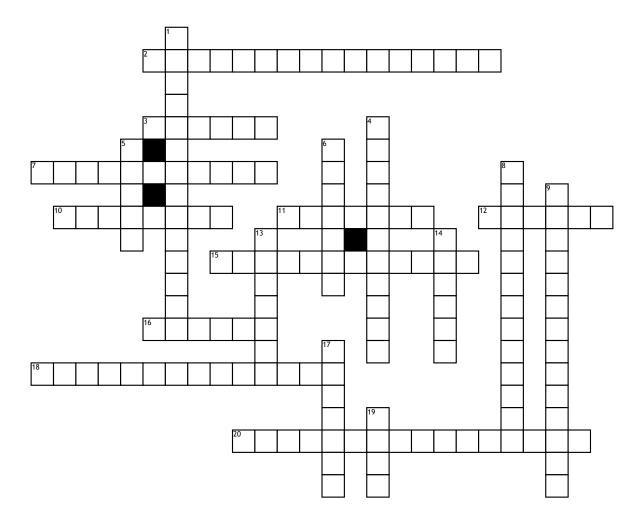
Name:	Date:	

## **Matter Review**



## **Across**

- **2.** a property of matter that describes a substance's ability to participate in chemical reactions
- 3. the state of matter that has a definite volume but not a definite shape7. the change of a substance from a liquid to a gas
- 10. the curve at a liquid's sur- face by which one measures the volume of the liquid
- 11. the change of state in which a solid becomes a liquid by adding energy
- **12.** anything that has mass and takes up space.
- **15.** the change of state from a gas to a liquid

- **16.** a measure of the gravita- tional force exerted on an object; its value can change with the location of the object in the universe
- **18.** a change that occurs when one or more substances change into entirely new substances with different properties
- **20.** a characteristic of a substance that does not involve a chemical change, such as density, color, or hardness

## **Down**

- 1. a change of mat- ter from one form to another without a change in chemical properties
- 4. the process in which a solid changes directly into a gas
- **5.** the state of matter in which the volume and shape of a substance are fixed

- **6.** the conversion of a liquid to a vapor when the vapor pressure of the liquid equals the atmospheric pressure
- **8.** the change of a substance from one physical state to another
- **9.** the physical forms of matter, which include solid, liquid, and gas
- **13.** the tendency of an object to resist being moved or, if the object is moving, to resist a change in speed or direction until an outside force acts on the object
- **14.** a measure of the size of a body or region in three-dimensional space
- 17. the ratio of the mass of a substance to the volume of the substance
- **19.** a measure of the amount of matter in an object