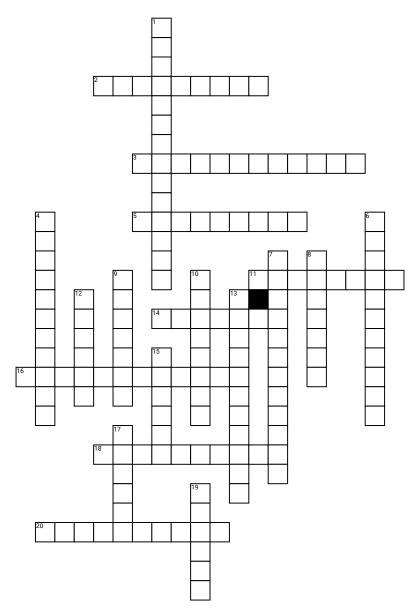
## **Mixtures**



## Across

- **2.** a substance that does not dissolve is said to be
- **3.** A solution where lots of solute has been added
- **5.** A solution that no more solute will dissolve in to
- **11.** The material that does not dissolve sinks to the bottom and forms a
- **14.** the substance that does the dissolving
- **16.** A method of separating a solvent (which is collected) from a soluble substance by heating

- **18.** A method of separating insoluble materials due to their size
- **20.** A solution where the particles settle after standing for a few minutes

## Down

- **1.** A method of separating soluble dyes
- **4.** A method of separating a solvent (which is not collected) from a soluble substance by heating
- **6.** How solutions look
- 7. method used to split blood in to its different parts

- **8.** two or more substances not chemically combined
- **9.** A substance that dissolves is said to be
- **10.** combination of a solute and a solvent
- **12.** the substance that is dissolved
- **13.** Method used to separate oil and water
- **15.** Iron, nickel and cobalt can be separated this way
- **17.** A solution where only a small amount of solute has been added
- **19.** When a solute doesn't dissolve the mixture will look